CPC COOPERATIVE PATENT CLASSIFICATION

C CHEMISTRY; METALLURGY

(NOTES omitted)

CHEMISTRY

C01 INORGANIC CHEMISTRY

(NOTES omitted)

C01B NON-METALLIC ELEMENTS; COMPOUNDS THEREOF; {METALLOIDS OR COMPOUNDS THEREOF NOT COVERED BY SUBCLASS C01C}

NOTES

- 1. In this subclass, tradenames that are often found in scientific and patent literature have been used in order to define precisely the scope of the groups.
- 2. Attention is drawn to the definitions of groups of chemical elements following the title of section C.

WARNINGS

1. The following IPC groups are not in the CPC scheme. The subject matter for these IPC groups is classified in the following CPC groups:

C01B 35/16, C01B 35/18 covered by <u>C01B 35/00</u>

2. In this subclass non-limiting references (in the sense of paragraph 39 of the Guide to the IPC) may still be displayed in the scheme

Hydrogen; Hydrides; Water; Synthesis gas from hydrocarbons

3/00 Hydrogen; Gaseous mixtures containing hydrogen; Separation of hydrogen from mixtures containing it (separation of gases by physical means B01D); Purification of hydrogen (production of water gas or synthesis gas from solid carbonaceous material C10J; purifying or modifying the chemical compositions of combustible technical gases containing carbon monoxide C10K)

NOTES

3/0021

- 1. In this group it is desirable to add the indexing codes of groups B01J 2208/00 and B01J 2219/00, for details relating to the reactors used in the generation of hydrogen or synthesis gas.
- In groups <u>C01B 3/12 C01B 3/18</u> and in groups <u>C01B 3/22 - C01B 3/586</u> it is desirable to add the indexing codes of group <u>C01B 2203/00</u>, for aspects relating to hydrogen or synthesis gas generation processes.

3/0005 • {Reversible uptake of hydrogen by an appropriate medium, i.e. based on physical or chemical sorption phenomena or on reversible chemical reactions, e.g. for hydrogen storage purposes (purification of hydrogen C01B 3/508); Reversible gettering of hydrogen; Reversible uptake of hydrogen by electrodes}

3/001 • Characterised by the uptaking medium;
Treatment thereof}

3/0015 . . . {Organic compounds; Solutions thereof}

• • • {Carbon, e.g. active carbon, carbon nanotubes, fullerenes; Treatment thereof}

3/0026 . . . { of one single metal or a rare earth metal; Treatment thereof}

NOTES

- In all of the groups
 <u>C01B 3/0026</u> <u>C01B 3/0084</u>, the metallic
 storage materials may contain minor
 quantities of non-metals such as B, C,
 O, S, Se, Si; e.g. <u>C01B 3/0036</u> "only
 containing iron and titanium" includes Fe-Ti
 compositions comprising non-metals
- 2. In the groups C01B 3/0026 and C01B 3/0047 C01B 3/0068 a "rareearth metal" means one single metal or a combination of metals selected from the lanthanides, Sc or Y

3/0031 . . . {Intermetallic compounds; Metal alloys; Treatment thereof}

3/0036 {only containing iron and titanium; Treatment thereof}

3/0042 {only containing magnesium and nickel; Treatment thereof}

3/0047 {containing a rare earth metal; Treatment thereof}

3/0052 {also containing titanium} 3/0057 {also containing nickel}

3/0063 {only containing a rare earth metal and only one other metal}

3/0068 {the other metal being nickel}

3/0073 . . . {Slurries, Suspensions}

2/0070			
3/0078	• • • {Composite solid storage mediums, i.e.	3/346	• • • {using heat generated by superheated steam}
	coherent or loose mixtures of different solid	3/348	• • • {by direct contact with heat accumulating
	constituents, chemically or structurally		liquids, e.g. molten metals, molten salts}
	heterogeneous solid masses, coated solids or	3/36	using oxygen or mixtures containing oxygen
	solids having a chemically modified surface		as gasifying agents
2/0004	region}	3/363	• • • • {characterised by the burner used}
3/0084	• • • {Solid storage mediums characterised by their	3/366	{Partial combustion in internal-
	shape, e.g. pellets, sintered shaped bodies, sheets, porous compacts, spongy metals,		combustion engines}
	hollow particles, solids with cavities, layered	3/38	using catalysts
	solids}	3/382	• • • • {Multi-step processes}
3/0089	• {Ortho-para conversion}	3/384	{the catalyst being continuously externally
3/0094	• {Atomic hydrogen}	0.000	heated}
3/02	• Production of hydrogen or of gaseous mixtures	3/386	{Catalytic partial combustion}
	containing {a substantial proportion of} hydrogen	3/388	• • • • {the heat being generated by superheated
3/025	• • {Preparation or purification of gas mixtures for	2/40	steam}
	ammonia synthesis}	3/40	characterised by the catalyst
3/04	by decomposition of inorganic compounds, e.g.	3/42	using moving solid particles
	ammonia {(C01B 3/0005 takes precedence)}	3/44	using the fluidised bed technique
3/042	• • • {Decomposition of water}	3/46	using discontinuously preheated non-moving
3/045	• • • {in gaseous phase}	2/49	solid materials, e.g. blast and run
3/047	{Decomposition of ammonia}	3/48	followed by reaction of water vapour with carbon monoxide
3/06	by reaction of inorganic compounds containing	3/50	Separation of hydrogen or hydrogen containing
	electro-positively bound hydrogen, e.g. water,	3/30	gases from gaseous mixtures, e.g. purification
	acids, bases, ammonia, with inorganic reducing		(C01B 3/14 takes precedence)
	agents (by electrolysis of water <u>C25B 1/04</u>)	3/501	• • {by diffusion}
3/061	• • • {by reaction of metal oxides with water}	3/503	• • (characterised by the membrane)
3/063	{Cyclic methods}	3/505	{Membranes containing palladium}
3/065	• • • {from a hydride}	3/506	• • {at low temperatures}
3/066	• • • {by reaction of water with phosphorus}	3/508	• • {at low temperatures} • • {by selective and reversible uptake by an
3/068	• • • {the hydrogen being generated from the water	3/308	appropriate medium, i.e. the uptake being based
	as a result of a cyclus of reactions, not covered		on physical or chemical sorption phenomena or
	by groups <u>C01B 3/063</u> or <u>C01B 3/105</u> }		on reversible chemical reactions (the appropriate
3/08	with metals		mediums per se C01B 3/0005)}
3/10	by reaction of water vapour with metals	3/52	by contacting with liquids; Regeneration of used
3/105	{Cyclic methods}	3/52	• • by contacting with liquids; Regeneration of used liquids {(C01B 3/508 takes precedence)}
	 {Cyclic methods} by reaction of water vapour with carbon	3/52 3/54	
3/105 3/12	 {Cyclic methods} by reaction of water vapour with carbon monoxide		liquids {(C01B 3/508 takes precedence)} including a catalytic reaction . by contacting with solids; Regeneration of used
3/105 3/12 3/14	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam 	3/54 3/56	liquids {(C01B 3/508 takes precedence)} including a catalytic reaction
3/105 3/12 3/14 3/16	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts 	3/54	liquids {(C01B 3/508 takes precedence)} including a catalytic reaction . by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction
3/105 3/12 3/14 3/16 3/18	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts using moving solid particles 	3/54 3/56	liquids {(C01B 3/508 takes precedence)} including a catalytic reaction . by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of
3/105 3/12 3/14 3/16	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts using moving solid particles by reaction of metal hydroxides with carbon 	3/54 3/56 3/58	liquids {(C01B 3/508 takes precedence)} including a catalytic reaction . by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide}
3/105 3/12 3/14 3/16 3/18 3/20	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts using moving solid particles by reaction of metal hydroxides with carbon monoxide 	3/54 3/56 3/58	liquids {(C01B 3/508 takes precedence)} including a catalytic reaction . by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of
3/105 3/12 3/14 3/16 3/18	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts using moving solid particles by reaction of metal hydroxides with carbon monoxide by decomposition of gaseous or liquid 	3/54 3/56 3/58 3/583 3/586	liquids {(C01B 3/508 takes precedence)} including a catalytic reaction . by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction}
3/105 3/12 3/14 3/16 3/18 3/20	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts using moving solid particles by reaction of metal hydroxides with carbon monoxide . by decomposition of gaseous or liquid organic compounds ({C01B 3/0005 takes}) 	3/54 3/56 3/58 3/583	liquids {(C01B 3/508 takes precedence)} including a catalytic reaction . by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof
3/105 3/12 3/14 3/16 3/18 3/20	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts using moving solid particles by reaction of metal hydroxides with carbon monoxide . by decomposition of gaseous or liquid organic compounds ({C01B 3/0005 takes precedence}); coking liquid carbonaceous 	3/54 3/56 3/58 3/583 3/586	 liquids {(C01B 3/508 takes precedence)} including a catalytic reaction by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ →
3/105 3/12 3/14 3/16 3/18 3/20 3/22	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts using moving solid particles by reaction of metal hydroxides with carbon monoxide . by decomposition of gaseous or liquid organic compounds ({C01B 3/0005 takes precedence}; coking liquid carbonaceous materials C10B 55/00) 	3/54 3/56 3/58 3/583 3/586 4/00	 liquids {(C01B 3/508 takes precedence)} including a catalytic reaction by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD
3/105 3/12 3/14 3/16 3/18 3/20 3/22	 {Cyclic methods} by reaction of water vapour with carbon monoxide 	3/54 3/56 3/58 3/583 3/586 4/00 5/00	 liquids {(C01B 3/508 takes precedence)} including a catalytic reaction by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD Water
3/105 3/12 3/14 3/16 3/18 3/20 3/22 3/24 3/26	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts using moving solid particles by reaction of metal hydroxides with carbon monoxide . by decomposition of gaseous or liquid organic compounds ({C01B 3/0005 takes precedence } ; coking liquid carbonaceous materials C10B 55/00) . of hydrocarbons . using catalysts 	3/54 3/56 3/58 3/583 3/586 4/00	 liquids {(C01B 3/508 takes precedence)} including a catalytic reaction by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD Water Heavy water; Preparation by chemical reaction of
3/105 3/12 3/14 3/16 3/18 3/20 3/22 3/22 3/24 3/26 3/28	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts using moving solid particles by reaction of metal hydroxides with carbon monoxide . by decomposition of gaseous or liquid organic compounds ({C01B 3/0005 takes precedence } ; coking liquid carbonaceous materials C10B 55/00) of hydrocarbons using catalysts using moving solid particles 	3/54 3/56 3/58 3/583 3/586 4/00 5/00	 liquids {(C01B 3/508 takes precedence)} including a catalytic reaction by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD Water Heavy water; Preparation by chemical reaction of hydrogen isotopes or their compounds, e.g. 4ND₃ +
3/105 3/12 3/14 3/16 3/18 3/20 3/22 3/24 3/26 3/28 3/30	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts using moving solid particles by reaction of metal hydroxides with carbon monoxide . by decomposition of gaseous or liquid organic compounds ({C01B 3/0005 takes precedence} ; coking liquid carbonaceous materials C10B 55/00) of hydrocarbons using catalysts using moving solid particles using the fluidised bed technique 	3/54 3/56 3/58 3/583 3/586 4/00 5/00	 liquids {(C01B 3/508 takes precedence)} including a catalytic reaction by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD Water Heavy water; Preparation by chemical reaction of
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3/105 3/12 3/14 3/16 3/18 3/20 3/22 3/24 3/26 3/28 3/30 3/32	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts using moving solid particles by reaction of metal hydroxides with carbon monoxide . by decomposition of gaseous or liquid organic compounds ({C01B 3/0005 takes precedence}; coking liquid carbonaceous materials C10B 55/00) . of hydrocarbons using catalysts using moving solid particles using the fluidised bed technique . by reaction of gaseous or liquid organic compounds with gasifying agents, e.g. water, 	3/54 3/56 3/58 3/583 3/586 4/00 5/02	 liquids {(C01B 3/508 takes precedence)} . including a catalytic reaction . by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} . including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD Water . Heavy water; Preparation by chemical reaction of hydrogen isotopes or their compounds, e.g. 4ND₃ + 7O₂> 4NO₂ + 6D₂O, 2D₂ + O₂> 2D₂O Hydrides of metals {including fully or partially hydrided metals, alloys or intermetallic compounds (use of some thereof for reversible sorption of hydrogen C01B 3/0005, C01B 3/508);
3/105 3/12 3/14 3/16 3/18 3/20 3/22 3/24 3/26 3/28 3/30 3/32	 {Cyclic methods} by reaction of water vapour with carbon monoxide Handling of heat and steam using catalysts using moving solid particles by reaction of metal hydroxides with carbon monoxide . by decomposition of gaseous or liquid organic compounds ({C01B 3/0005 takes precedence } ; coking liquid carbonaceous materials C10B 55/00) . of hydrocarbons using catalysts using moving solid particles using the fluidised bed technique . by reaction of gaseous or liquid organic compounds with gasifying agents, e.g. water, carbon dioxide, air {Catalytic reaction of gaseous or liquid organic 	3/54 3/56 3/58 3/583 3/586 4/00 5/02	 liquids {(C01B 3/508 takes precedence)} . including a catalytic reaction . by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} . including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD Water . Heavy water; Preparation by chemical reaction of hydrogen isotopes or their compounds, e.g. 4ND₃ + 7O₂> 4NO₂ + 6D₂O, 2D₂ + O₂> 2D₂O Hydrides of metals {including fully or partially hydrided metals, alloys or intermetallic compounds (use of some thereof for reversible sorption of hydrogen C01B 3/0005, C01B 3/508); Compounds containing at least one metal-
3/105 3/12 3/14 3/16 3/18 3/20 3/22 3/24 3/26 3/28 3/30 3/32	 {Cyclic methods} by reaction of water vapour with carbon monoxide 	3/54 3/56 3/58 3/583 3/586 4/00 5/02	 liquids {(C01B 3/508 takes precedence)} . including a catalytic reaction . by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} . including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD Water . Heavy water; Preparation by chemical reaction of hydrogen isotopes or their compounds, e.g. 4ND₃ + 7O₂> 4NO₂ + 6D₂O, 2D₂ + O₂> 2D₂O Hydrides of metals {including fully or partially hydrided metals, alloys or intermetallic compounds (use of some thereof for reversible sorption of hydrogen C01B 3/0005, C01B 3/508); Compounds containing at least one metal-hydrogen bond, e.g. (GeH₃)₂S, SiH GeH};
3/105 3/12 3/14 3/16 3/18 3/20 3/22 3/24 3/26 3/28 3/30 3/32 3/323	 {Cyclic methods} by reaction of water vapour with carbon monoxide 	3/54 3/56 3/58 3/583 3/586 4/00 5/02	liquids {(C01B 3/508 takes precedence)} including a catalytic reaction by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD Water . Heavy water; Preparation by chemical reaction of hydrogen isotopes or their compounds, e.g. 4ND₃ + 7O₂> 4NO₂ + 6D₂O, 2D₂ + O₂> 2D₂O Hydrides of metals {including fully or partially hydrided metals, alloys or intermetallic compounds (use of some thereof for reversible sorption of hydrogen C01B 3/0005, C01B 3/508); Compounds containing at least one metalhydrogen bond, e.g. (GeH₃)₂S, SiH GeH}; Monoborane or diborane; Addition complexes
3/105 3/12 3/14 3/16 3/18 3/20 3/22 3/22 3/24 3/26 3/28 3/30 3/32 3/323	 {Cyclic methods} by reaction of water vapour with carbon monoxide 	3/54 3/56 3/58 3/583 3/586 4/00 5/02	liquids {(C01B 3/508 takes precedence)} including a catalytic reaction by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD Water . Heavy water; Preparation by chemical reaction of hydrogen isotopes or their compounds, e.g. 4ND₃ + 7O₂> 4NO₂ + 6D₂O, 2D₂ + O₂> 2D₂O Hydrides of metals {including fully or partially hydrided metals, alloys or intermetallic compounds (use of some thereof for reversible sorption of hydrogen C01B 3/0005, C01B 3/508); Compounds containing at least one metalhydrogen bond, e.g. (GeH₃)₂S, SiH GeH}; Monoborane or diborane; Addition complexes thereof (higher hydrides of boron, substituted
3/105 3/12 3/14 3/16 3/18 3/20 3/22 3/22 3/24 3/26 3/28 3/30 3/32 3/323	 {Cyclic methods} by reaction of water vapour with carbon monoxide 	3/54 3/56 3/58 3/583 3/586 4/00 5/02 6/00	liquids {(C01B 3/508 takes precedence)} including a catalytic reaction by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD Water . Heavy water; Preparation by chemical reaction of hydrogen isotopes or their compounds, e.g. 4ND₃ + 7O₂> 4NO₂ + 6D₂O, 2D₂ + O₂> 2D₂O Hydrides of metals {including fully or partially hydrided metals, alloys or intermetallic compounds (use of some thereof for reversible sorption of hydrogen C01B 3/0005, C01B 3/508); Compounds containing at least one metalhydrogen bond, e.g. (GeH₃)₂S, SiH GeH}; Monoborane or diborane; Addition complexes thereof (higher hydrides of boron, substituted hydrides of boron C01B 35/00)
3/105 3/12 3/14 3/16 3/18 3/20 3/22 3/22 3/24 3/26 3/28 3/30 3/32 3/323	 {Cyclic methods} by reaction of water vapour with carbon monoxide 	3/54 3/56 3/58 3/583 3/586 4/00 5/02	liquids {(C01B 3/508 takes precedence)} including a catalytic reaction by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD Water . Heavy water; Preparation by chemical reaction of hydrogen isotopes or their compounds, e.g. 4ND₃ + 7O₂> 4NO₂ + 6D₂O, 2D₂ + O₂> 2D₂O Hydrides of metals {including fully or partially hydrided metals, alloys or intermetallic compounds (use of some thereof for reversible sorption of hydrogen C01B 3/0005, C01B 3/508); Compounds containing at least one metalhydrogen bond, e.g. (GeH₃)₂S, SiH GeH}; Monoborane or diborane; Addition complexes thereof (higher hydrides of boron, substituted hydrides of boron C01B 35/00) . {Hydrides containing only one metal and one or
3/105 3/12 3/14 3/16 3/18 3/20 3/22 3/22 3/24 3/26 3/28 3/30 3/32 3/323	 {Cyclic methods} by reaction of water vapour with carbon monoxide 	3/54 3/56 3/58 3/583 3/586 4/00 5/02 6/00	liquids {(C01B 3/508 takes precedence)} including a catalytic reaction by contacting with solids; Regeneration of used solids {(C01B 3/508 takes precedence)} including a catalytic reaction {the reaction being the selective oxidation of carbon monoxide} {the reaction being a methanation reaction} Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. NH₃ + D₂ → NH₂D + HD Water . Heavy water; Preparation by chemical reaction of hydrogen isotopes or their compounds, e.g. 4ND₃ + 7O₂> 4NO₂ + 6D₂O, 2D₂ + O₂> 2D₂O Hydrides of metals {including fully or partially hydrided metals, alloys or intermetallic compounds (use of some thereof for reversible sorption of hydrogen C01B 3/0005, C01B 3/508); Compounds containing at least one metalhydrogen bond, e.g. (GeH₃)₂S, SiH GeH}; Monoborane or diborane; Addition complexes thereof (higher hydrides of boron, substituted hydrides of boron C01B 35/00)

6/02	• Hydrides of transition elements; Addition	7/0718	• • • {by adsorption}
	complexes thereof	7/0725	• • • • {by active carbon}
6/04	 Hydrides of alkali metals, alkaline earth metals, 	7/0731	• • • {by extraction}
	beryllium or magnesium; Addition complexes	7/0737	{hydrogen chloride being extracted}
	thereof	7/0743	• • · {of gaseous or dissolved chlorine}
6/06	. Hydrides of aluminium, gallium, indium, thallium,	7/075	• • • of liquid chlorine
	germanium, tin, lead, arsenic, antimony, bismuth	7/073	Bromine; Hydrogen bromide
	or polonium; Monoborane; Diborane; Addition		· -
	complexes thereof	7/093	• • {Hydrogen bromide}
6/065	{Hydrides of arsenic or antimony}	7/096	• • {Bromine}
6/10	Monoborane; Diborane; Addition complexes	7/13	Iodine; Hydrogen iodide
2. 2 2	thereof	7/135	• • {Hydrogen iodide}
6/11	Preparation from boron or inorganic	7/14	Iodine
0/11	compounds containing boron and oxygen	7/16	Preparation from seaweed
6/13	Addition complexes of monoborane or	7/19	Fluorine; Hydrogen fluoride
0/13	diborane, e.g. with phosphine, arsine or	7/191	• • {Hydrogen fluoride}
	hydrazine	7/192	{Preparation from fluorspar}
6/15	Metal borohydrides; Addition complexes	7/193	• • • {Preparation from silicon tetrafluoride,
0/13	thereof	,,1,5	fluosilicic acid or fluosilicates}
6/17	Preparation from boron or inorganic	7/194	• • • {Preparation from ammonium fluoride}
6/17	compounds containing boron and oxygen	7/195	{Separation; Purification}
6/10		7/195	
6/19	Preparation from other compounds of		• • • {by distillation}
- 10 -	boron	7/197	• • • {by adsorption}
6/21	Preparation of borohydrides of	7/198	• • • • {by solid ion-exchangers}
	alkali metals, alkaline earth metals,	7/20	Fluorine
	magnesium or beryllium; Addition	7/24	 Inter-halogen compounds
	complexes thereof, e.g. LiBH _{4.2} N ₂ H ₄ ,	9/00	General methods of preparing halides (particular
< 12.2	NaB ₂ H ₇	2/00	individual halides, see the relevant groups in
6/23	Preparation of borohydrides of other		C01B - C01G according to the element combined
	metals, e.g. aluminium borohydride;		with the halogen; electrolytic production of inorganic
	Addition complexes thereof, e.g.		
- 12.1	Li[Al(BH ₄) ₃ H]	0/02	compounds <u>C25B</u>)
6/24	• Hydrides containing at least two metals; Addition	9/02	. Chlorides
	complexes thereof (C01B 6/13 - C01B 6/23 take	9/04	. Bromides
	precedence)	9/06	. Iodides
6/243	• • {containing only hydrogen, aluminium and alkali	9/08	• Fluorides
	metals, e.g. Li(AlH ₄)}	11/00	Oxides or oxyacids of halogens; Salts thereof
6/246	• • {also containing non-metals other than hydrogen}	11/02	• Oxides of chlorine
6/26	Preparation from the metal with the highest	11/021	• Chlorine hemioxide (Cl ₂ O)}
	valency or from its oxides or salts of its oxyacids	11/021	. {Chlorine dioxide (ClO ₂)}
6/34	 Purification; Stabilisation 	11/022	
			{Preparation from chlorites or chlorates}
Halogens, co	mpounds thereof	11/024	• • • • {from chlorites}
7/00	Halogens; Halogen acids (oxyacids C01B 11/00)	11/025	• • • • (from chlorates without any other reaction
7/01	. Chlorine; Hydrogen chloride		reducing agent than chloride ions}
7/012	Preparation of hydrogen chloride from the	11/026	• • • • {from chlorate ions in the presence of a
7/012	elements		peroxidic compound, e.g. hydrogen peroxide,
7/015	. • {Chlorine hydrates; Obtaining chlorine		ozone, peroxysulfates}
7/013	therefrom	11/027	• • • • {from chlorate ions in the presence of
7/017	Preparation of hydrogen chloride by reacting		a nitrogen compound selected from
7/017	together chlorine, water and carbon or carbon		nitrogen dioxide, nitrate or nitrite ions,
	monoxide (the carbon not acting only as		nitrosylchloride, hydrazine or hydrazine
			compounds}
7/02	catalyst)}	11/028	{Separation; Purification}
7/03	. Preparation from chlorides	11/029	• • {Chlorine trioxide (ClO ₃); Chlorine hexoxide
7/035	• • • {Preparation of hydrogen chloride from		(Cl_2O_6) ; Chlorine heptoxide (Cl_2O_7) }
5 /0 :	chlorides}	11/04	. Hypochlorous acid
7/04	• • Preparation of chlorine from hydrogen chloride	11/06	Hypochlorites
7/05	Preparation from ammonium chloride	11/062	• • • {Hypochlorites of alkali metals}
7/055	• • • • {Preparation of hydrogen chloride from	11/064	{Hypochlorites of alkaline-earth metals}
	ammonium chloride}	11/066	{Magnesium hypochlorite}
7/07	• • Purification {; Separation (C01B 7/015 takes	11/068	• • • {Stabilisation by additives other than oxides,
	precedence)}		hydroxides, carbonates of alkali or alkaline-
7/0706	• • • {of hydrogen chloride}		earth metals; Coating of particles; Shaping;
7/0712	• • • {by distillation}		Granulation}

11/08	Chlorous acid	13/08	• • from air with the aid of metal oxides, e.g. barium
11/10	Chlorites		oxide, manganese oxide {(C01B 13/0292 takes
11/12	Chloric acid		precedence)}
11/14	Chlorates	13/083	• • { with barium oxide }
11/145	{Separation; Crystallisation; Purification,	13/086	• • { with manganese oxide }
	After-treatment; Stabilisation by additives}	13/10	 Preparation of ozone
11/16	Perchloric acid	13/11	• • by electric discharge
11/18	Perchlorates		NOTE
11/185	• • {Ammonium perchlorate}		
11/20	Oxygen compounds of bromine		In groups <u>C01B 13/11</u> and <u>C01B 13/115</u> ,
11/22	• Oxygen compounds of iodine		additional features relating to the
11/24	Oxygen compounds of fluorine		preparation of ozone by electrical discharge
11/27	• Oxygen compounds of fluorine		are indexed with codes chosen from
Oxygen; Oxio	des or hydroxides in general; Per-compounds		<u>C01B 2201/00</u> - <u>C01B 2201/90</u> .
		13/115	{characterised by the electrical circuits
13/00	Oxygen; Ozone; Oxides or hydroxides in general		producing the electrical discharge}
13/02	• Preparation of oxygen (by liquefying <u>F25J</u>)	13/14	 Methods for preparing oxides or hydroxides in
13/0203	• • {from inorganic compounds}		general (particular individual oxides or hydroxides,
13/0207	• • {Water}		see the relevant groups of subclasses C01B - C01G
13/0211	• • {Peroxy compounds}		or C25B, according to the element combined with
13/0214	• • • {Hydrogen peroxide}		the oxygen or hydroxy group)
13/0218	{Chlorate}	13/145	• • {After-treatment of oxides or hydroxides, e.g.
13/0222	• • {from organic compounds}		pulverising, drying, decreasing the acidity}
13/0225	• • • {Peroxy compounds}	13/16	Purification
13/0229	• • {Purification or separation processes}	13/18	by thermal decomposition of compounds, e.g. of
	NOTE		salts or hydroxides
	NOTE	13/185	• • • {Preparing mixtures of oxides}
	In groups <u>C01B 13/0229</u> - <u>C01B 13/0288</u> ,	13/20	by oxidation of elements in the gaseous state;
	additional features relating to the		by oxidation or hydrolysis of compounds in the
	purification or separation processes		gaseous state
	are indexed with codes chosen from	13/22	of halides or oxyhalides
	<u>C01B 2210/0026</u> - <u>C01B 2210/0098</u> .	13/24	in the presence of hot combustion gases
13/0233	• • {Chemical processing only}	13/26	in the presence of a fluidised bed
13/0237	• • • {by oxidation}	13/28	using a plasma or an electric discharge
13/024	{by reduction}	13/30	Removal and cooling of the oxide-containing
13/0244	{by complexation}		suspension
13/0248	{Physical processing only}	13/32	by oxidation or hydrolysis of elements or
13/0248	• • • {I hysical processing only} • • • • {by making use of membranes}		compounds in the liquid or solid state {or in non-
13/0251	• • • • {characterised by the type of membrane}		aqueous solution, e.g. sol-gel process}
		13/322	{of elements or compounds in the solid state}
13/0259	• • • {by adsorption on solids}	13/324	• • • {by solid combustion synthesis}
13/0262	{characterised by the adsorbent}	13/326	• • { of elements or compounds in the liquid state }
13/0266	{Carbon based materials}	13/328	{by processes making use of emulsions, e.g. the
13/027	{Zeolites}		kerosine process}
13/0274	{Other molecular sieve materials}	13/34	by oxidation or hydrolysis of sprayed or atomised
13/0277	{Temperature swing adsorption}		solutions
13/0281	· · · · · {in getters}	13/36	• • by precipitation reactions in {aqueous} solutions
13/0285	• • • {by absorption in liquids}		$\{(C01B 13/328 \text{ takes precedence})\}$
13/0288	• • • {Combined chemical and physical processing}	13/363	{Mixtures of oxides or hydroxides by
	NOTE		precipitation}
	I- 4hi	13/366	• • {by hydrothermal processing}
	In this group, processing steps are indexed with codes chosen from		• • •
	C01B 2210/0001 - C01B 2210/0025	15/00	Peroxides; Peroxyhydrates; Peroxyacids or salts
	<u>COTB 2210/0001</u> - <u>COTB 2210/0025</u>	4.5.00.5	thereof; Superoxides; Ozonides
13/0292	• • {Preparation from air using a molten phase	15/005	• {Stabilisation of the solid compounds subsequent to
	containing alkali metal nitrite, optionally together		the preparation or to the crystallisation, by additives
	with other oxygen acceptors}	15/01	or by coating}
13/0296	• • {Generators releasing in a self-sustaining	15/01	. Hydrogen peroxide
	way pure oxygen from a solid charge, without	15/013	Separation; Purification; Concentration
	interaction of it with a fluid nor external heating,	15/0135	• • • {Purification by solid ion-exchangers or solid
	e.g. chlorate candles or canisters containing them	15/015	chelating agents}
	(charges <u>per se</u> <u>C01B 13/02</u>)}	15/017	Anhydrous hydrogen peroxide; Anhydrous
			solutions or gaseous mixtures containing
			hydrogen peroxide

15/022	Preparation from organic compounds	17/033	using a liquid extractant
15/023	by the alkyl-anthraquinone process	17/04	from gaseous sulfur compounds including
15/024	· · · from hydrocarbons	17,01	gaseous sulfides
15/026	from alcohols	17/0404	• • {by processes comprising a dry catalytic
15/027	Preparation from water		conversion of hydrogen sulfide-containing
15/0275	• • • {Preparation by reaction of water, carbon		gases, e.g. the Claus process}
	monoxide and oxygen}	17/0408	• • • {Pretreatment of the hydrogen sulfide containing gases}
15/029	Preparation from hydrogen and oxygen	17/0413	{characterised by the combustion step}
15/0295	• • {by electrical discharge}	17/0417	{Combustion reactors}
15/03	• Preparation from inorganic peroxy compounds,	17/0417	{Multistage combustion}
15/022	e.g. from peroxysulfates	17/0421	{characterised by the catalytic conversion}
15/032	from metal peroxides	17/0420	{Catalytic converters}
15/037 15/04	 Stabilisation by additives Metal peroxides or peroxyhydrates thereof; {Metal} 	17/0434	{Catalyst compositions}
13/04	superoxides; {Metal} ozonides; {Peroxyhydrates thereof}	17/0439	• • • • {at least one catalyst bed operating below the dew-point of sulfur}
15/043	• of alkali metals, alkaline earth metals or	17/0443	· · · · { (in a moving bed}
13/043	magnesium {or beryllium or aluminium}	17/0447	{Separation of the obtained sulfur}
15/0435	• • { of alkali metals }	17/0452	{Process control; Start-up or cooling-down
15/047	. of heavy metals	17/0432	procedures of the Claus process}
15/0475	• • {of actinides}	17/0456	• • • • {the hydrogen sulfide-containing gas being a
15/055	Peroxyhydrates (C01B 15/04 takes precedence);	1770130	Claus process tail gas}
13/033	Peroxyacids or salts thereof	17/046	• • • • { without intermediate formation of sulfur
15/06	containing sulfur		dioxide}
15/08	Peroxysulfates	17/0465	• • • • {Catalyst compositions}
15/085	• • • {Stabilisation of the solid compounds,	17/0469	• • • • {at least one catalyst bed operating below
10,000	subsequent to the preparation or to the		the dew-point of sulfur}
	crystallisation, by additives or by coating}	17/0473	• • • {by reaction of sulfur dioxide or sulfur trioxide
15/10	containing carbon		containing gases with reducing agents other
15/103	{containing only alkali metals as metals}		than hydrogen sulfide}
15/106	{Stabilisation of the solid compounds,	17/0478	• • • { with hydrocarbons or mixtures containing
	subsequent to the preparation or to the		them}
	crystallisation, by additives or by coating}	17/0482	• • • { with carbon or solid carbonaceous
15/12	• containing boron		materials}
15/123	• • • {Stabilisation of the solid compounds,	17/0486	• • • {with carbon monoxide or carbon monoxide
	subsequent to the preparation or to the	17/0401	containing mixtures}
	crystallisation, by additives or by coating}	17/0491	• • • { with hydrogen or hydrogen-containing
15/126	• • • {Dehydration of solid hydrated peroxyborates	17/0495	mixtures, e.g. synthesis gas} {by dissociation of hydrogen sulfide into the
15/14	to less hydrated or anhydrous products}	17/0493	elements}
15/14	• containing silicon	17/05	by wet processes
15/16	containing phosphorus	17/05	 by wet processes from non-gaseous sulfides or materials containing
17/00	Sulfur; Compounds thereof	17700	such sulfides, e.g. ores
17/02	Preparation of sulfur; Purification	17/10	Finely divided sulfur, e.g. sublimed sulfur,
17/0205	• • {Separation of sulfur from liquids, e.g. by	17,10	flowers of sulfur
	coalescence}	17/12	Insoluble sulfur (mu-sulfur)
17/021	• • {Separation of sulfur from gases}	17/125	• • {Sulfur isotopes other than 32S}
17/0216	• • {Solidification or cooling of liquid sulfur}	17/16	Hydrogen sulfides
17/0221	{Melting}	17/161	• • {Preparation from elemental sulfur}
17/0226	• • {Vaporising or superheating}	17/162	• • • {from elemental sulfur and hydrogen}
17/0232	• • {Purification, e.g. degassing}	17/164	• • {Preparation by reduction of oxidic sulfur
17/0237	 {Converting into particles, e.g. by granulation, milling} 	17/165	compounds} . • {Preparation from sulfides, oxysulfides or
17/0243	• • {Other after-treatment of sulfur}	17/103	polysulfides}
17/0248	• • {of particulate sulfur}	17/167	• {Separation}
17/0253	• • (or particulate surfur) • • (from non-gaseous sulfur compounds other than	17/168	• {Separation} • • {Purification}
0 2 00	sulfides or materials containing such sulfides}	17/18	Hydrogen polysulfides
17/0259	{by reduction of sulfates}	17/10	 Methods for preparing sulfides or polysulfides,
17/0264	{of calcium sulfates}	11/20	in general (ammonium sulfides or polysulfides
17/027	Recovery of sulfur from material containing		C01C; sulfides or polysulfides of metals, other than
	elemental sulfur, e.g. luxmasses {or sulfur		alkali metals, magnesium, calcium, strontium and
	containing ores); Purification {of the recovered		barium, see the relevant groups of subclasses C01F
	sulfur}		or <u>C01G</u> , according to the metal)

17/22	 Alkali metal sulfides or polysulfides 	17/665	• • • {Stabilisation by additives subsequent to
17/24	Preparation by reduction		preparation; Dust prevention by additives}
17/26	• • • with carbon	17/69	Sulfur trioxide; Sulfuric acid
17/28	• • • with reducing gases	17/70	Stabilisation of gamma-form sulfur trioxide
17/30	Preparation from sodium or potassium amalgam	17/74	Preparation
	with sulfur or sulfides	17/745	• • {from sulfates}
17/32	Hydrosulfides of sodium or potassium	17/76	by contact processes
17/34	Polysulfides of sodium or potassium	17/762	• • • {High pressure processes}
17/36	• • Purification	17/765	Multi-stage SO ₃ -conversion
17/38	. Dehydration	17/7655	• • • • • {with intermediate absorption}
17/40	Making shaped products, e.g. granules	17/77	Fluidised-bed processes
17/42	 Sulfides or polysulfides of magnesium, calcium, 	17/775	Liquid phase contacting processes or wet
17/42	strontium, or barium	17/773	
17/42		17/70	catalysis processes
17/43	• • from oxides or hydroxides with sulfur or	17/78	• • • characterised by the catalyst used
17/44	hydrogen sulfide	17/79	containing vanadium
17/44	by reduction of sulfates	17/80	Apparatus
17/45	Compounds containing sulfur and halogen, with or	17/803	· · · · · {Converters}
	without oxygen	17/806	• • • • {Absorbers; Heat exchangers}
17/4507	• • {containing sulfur and halogen only}	17/82	of sulfuric acid using a nitrogen oxide process
17/4515	• • • {containing sulfur and fluorine only}	17/84	Chamber process
17/4523	• • • {Sulfur tetrafluoride}	17/86	Tower process
17/453	{Sulfur hexafluoride}	17/88	Concentration of sulfuric acid
17/4538	• • • {containing sulfur and chlorine only}	17/90	Separation; Purification
17/4546	{Sulfur dichloride}	17/901	• • • {Recovery from spent acids containing metallic
17/4553	{Sulfur hexachloride}	17701	ions, e.g. hydrolysis acids, pickling acids
17/4561	• • {Compounds containing sulfur, halogen and		(obtaining sulfur dioxide as an intermediate
177 1301	oxygen only}		in sulfur trioxide recovery from sulfates, e.g.
17/4569	• • • {Thionyl fluoride (SOF ₂)}		iron sulfates C01B 17/501, from spent acids
17/4576	{Sulfuryl fluoride (SO ₂ F ₂)}		<u>C01B 17/58</u>)}
		17/902	• • • • {by dialysis}
17/4584	{Thionyl chloride (SOCl ₂)}	17/903	{by liquid-liquid extraction}
17/4592	• • {Sulfuryl chloride (SO ₂ Cl ₂)}		
17/46	. Compounds containing sulfur, halogen, hydrogen,	17/904	{by ion-exchange}
	and oxygen	17/905	{Removal of organic impurities}
17/463	• • {Fluorosulfonic acid (FSO ₃ H)}	17/906	{Removal of mercury}
17/466	• • {Chlorosulfonic acid (ClSO ₃ H)}	17/907	• • • {Removal of arsenic}
17/48	 Sulfur dioxide; Sulfurous acid 	17/908	• • • {Removal of antimony or bismuth}
17/50	Preparation of sulfur dioxide	17/92	Recovery from acid tar or the like {, e.g.
17/501	• • • {by reduction of sulfur compounds}		alkylation acids (obtaining sulfur dioxide as
17/502	• • • {of sulfur trioxide}		an intermediate in sulfur trioxode recovery
17/503	• • • {of sulfuric acid}		therefrom <u>C01B 17/58</u>)}
17/504	• • • • {of ammonium sulfates (of ammonium	17/925	• • • {by processes involving a liquid-liquid
-,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	sulfates containing sulfuric acid solutions		extraction}
	<u>C01B 17/585</u>)}	17/94	Recovery from nitration acids
17/505	• • • {of alkali metal sulfates}	17/96	. Methods for the preparation of sulfates in general
17/506	{of calcium sulfates}		(particular individual sulfates, see the relevant
17/500	{of calcium surfaces}		groups of subclasses <u>C01B</u> - <u>C01G</u> , according to the
			cation)
17/508	• • • {by oxidation of sulfur compounds}	17/965	• • {Pyrosulfates}
17/52	by roasting sulfides (preliminary treatment of	17/98	Other compounds containing sulfur and oxygen
	ores or scrap <u>C22B 1/00</u>)		(persulfuric acids <u>C01B 15/06</u> ; persulfates
17/54	by burning elemental sulfur		C01B 15/08)
17/56	Separation; Purification		
17/58	Recovery of sulfur dioxide from acid tar or the	19/00	Selenium; Tellurium; Compounds thereof
	like {or from any waste sulfuric acid}		(phosphorus compounds <u>C01B 25/14</u>)
17/585	• • • • {from ammonium sulfate containing sulfuric	19/001	• {Preparation involving a liquid-liquid extraction, an
	acid solutions}		adsorption or an ion-exchange}
17/60	Isolation of sulfur dioxide from gases	19/002	 {Compounds containing, besides selenium or
17/62	• Methods of preparing sulfites in general (particular		tellurium, more than one other element, with -O-
	individual sulfites, see the relevant groups of		and -OH not being considered as anions}
	subclasses $\underline{\text{C01B}}$ - $\underline{\text{C01G}}$, according to the cation)	19/004	• {Oxides; Hydroxides}
17/625	• • {metabisulfites or pyrosulfites}	19/005	• {Halides}
17/64	Thiosulfates; Dithionites; Polythionates	19/007	• {Tellurides or selenides of metals (C01B 19/002
17/66	• Dithionites {or hydrosulfites (S ₂ O ₄ ²⁻)}		takes precedence)}

19/008	 {Salts of oxyacids of selenium or tellurium} 	21/0612 • • {with alkaline-earth metals, beryllium or
19/02	Elemental selenium or tellurium	magnesium}
19/04	 Binary compounds {including binary selenium- 	21/0615 • • { with transition metals other than titanium,
	tellurium compounds (<u>C01B 19/004</u> , <u>C01B 19/005</u> ,	zirconium or hafnium}
	<u>C01B 19/007</u> take precedence)}	21/0617 • • • { with vanadium, niobium or tantalum}
21/00	Nitrogon, Compounds though	21/062 • • • { with chromium, molybdenum or tungsten}
21/00	Nitrogen; Compounds thereof	21/0622 {with iron, cobalt or nickel}
21/02	• Preparation of nitrogen (by decomposition of	21/0625 {with copper}
21/04	ammonia { <u>C01B 3/047</u> })	21/0627 • • • {with one or more rare earth metals}
21/04	• Purification or separation of nitrogen (by liquefying	21/063 {with one or more actinides, e.g. UN, PuN}
21/2127	<u>F25J</u>)	21/0632 • • {with gallium, indium or thallium}
21/0405	• • {Purification or separation processes}	21/0635 •• {with garmanium, tin or lead}
	<u>NOTE</u>	21/0637 • • {with germanum, thi of leady}
	In this arrown additional factures relating	C01B 21/0607 - C01B 21/0635, other than
	In this group, additional features relating to the purification or separation processes	aluminium, titanium, zirconium or hafnium}
	are indexed with codes chosen from	21/064 • with boron
	C01B 2210/0026 - C01B 2210/0098	
	<u>COTB 2210/0020</u> - <u>COTB 2210/0098</u>	21/0641 {Preparation by direct nitridation of elemental boron}
21/0411	• • {Chemical processing only}	,
21/0416	• • • {by oxidation}	21/0643 {Preparation from boron halides}
21/0422	• • • {by reduction}	21/0645 {Preparation by carboreductive nitridation}
21/0427	{by complexation}	21/0646 {Preparation by pyrolysis of boron and nitrogen
21/0433	{Physical processing only}	containing compounds}
21/0438	{by making use of membranes}	21/0648 {After-treatment, e.g. grinding, purification
	` • • • • • • • • • • • • • • • • • • •	(transformation of hexagonal into cubic or
21/0444	{characterised by the membrane}	wurtzitic boron nitride <u>C04B 35/5831</u>)}
21/045	• • • {by adsorption in solids}	21/068 • • with silicon
21/0455	{characterised by the adsorbent}	21/0682 {Preparation by direct nitridation of silicon}
21/0461	• • • • • {Carbon based materials}	21/0685 {Preparation by carboreductive nitridation}
21/0466	· · · · · {Zeolites}	21/0687 • • • {After-treatment, e.g. grinding, purification}
21/0472	• • • • • {Other molecular sieve materials}	21/072 • • with aluminium
21/0477	• • • • {Temperature swing adsorption}	21/0722 {Preparation by direct nitridation of
21/0483	• • • • {in getters}	aluminium}
21/0488	• • • {by absorption in liquids}	
21/0400	• • • • {by absorption in fiquids}	21/0724 {using a plasma}
21/0488	{Obsorption in inquits} {Combined chemical and physical processing}	(2 1)
	{Combined chemical and physical processing}	21/0726 • • • {Preparation by carboreductive nitridation}
		21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification}
	• • • {Combined chemical and physical processing} NOTE In this group, processing steps are	21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium}
	{Combined chemical and physical processing} NOTE In this group, processing steps are indexed with codes chosen from	21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium,
	• • • {Combined chemical and physical processing} NOTE In this group, processing steps are	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium}
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or
	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides}
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides;	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation}
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal,	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08)	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal,	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds}
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification}
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES Binary compounds, i.e. compounds of nitrogen	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 . Hydrazoic acid; Azides; Halogen azides
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals,	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 {After-treatment, e.g. grinding, purification} 21/082 . Compounds containing nitrogen and non-metals
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 . Hydrazoic acid; Azides; Halogen azides 21/082 . Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076.	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 {After-treatment, e.g. grinding, purification} 21/08 Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence)
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 {After-treatment, e.g. grinding, purification} 21/08 {Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 . {Oxynitrides of metals, boron or silicon}
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 {After-treatment, e.g. grinding, purification} 21/08 {After-treatment, e.g. grinding, purification} 21/08 {Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 {Oxynitrides of metals, boron or silicon} 21/0823 {Silicon oxynitrides}
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one element chosen from metals, silicon or boron are classified in C01B 21/0602	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 {After-treatment, e.g. grinding, purification} 21/08 {After-treatment, e.g. grinding, purification} 21/08 {Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 {Oxynitrides of metals, boron or silicon} 21/0823 {Silicon oxynitrides} 21/0825 {Aluminium oxynitrides}
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one element chosen from metals, silicon or boron are	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 . Hydrazoic acid; Azides; Halogen azides 21/082 . Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0823 {Oxynitrides of metals, boron or silicon} 21/0825 {Aluminium oxynitrides} 21/0826 {Silicon aluminium oxynitrides, i.e. sialons}
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one element chosen from metals, silicon or boron are classified in C01B 21/0602 Documents relating to several specific binary	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 . Hydrazoic acid; Azides; Halogen azides 21/082 . Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 . {Oxynitrides of metals, boron or silicon} 21/0825 {Silicon oxynitrides} 21/0826 {Silicon aluminium oxynitrides, i.e. sialons} 21/0828 {Carbonitrides or oxycarbonitrides of metals,
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one element chosen from metals, silicon or boron are classified in C01B 21/0602 2. Documents relating to several specific binary compounds are classified in C01B 21/06 only	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 . Hydrazoic acid; Azides; Halogen azides 21/082 . Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 . {Oxynitrides of metals, boron or silicon} 21/0825 {Silicon oxynitrides} 21/0826 {Silicon aluminium oxynitrides, i.e. sialons} 21/0828 {Carbonitrides or oxycarbonitrides of metals, boron or silicon}
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one element chosen from metals, silicon or boron are classified in C01B 21/0602 2. Documents relating to several specific binary compounds are classified in C01B 21/06 only and receive the indexing codes chosen from	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 {After-treatment, e.g. grinding, purification} 21/08 {Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 {Oxynitrides of metals, boron or silicon} 21/0825 {Aluminium oxynitrides} 21/0826 {Silicon aluminium oxynitrides, i.e. sialons} 21/0828 {Carbonitrides or oxycarbonitrides of metals, boron or silicon} 21/083 containing one or more halogen atoms
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one element chosen from metals, silicon or boron are classified in C01B 21/0602 2. Documents relating to several specific binary compounds are classified in C01B 21/06 only and receive the indexing codes chosen from C01B 21/0602 - C01B 21/076 to identify the specific compounds	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 . Hydrazoic acid; Azides; Halogen azides 21/082 . Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 . {Oxynitrides of metals, boron or silicon} 21/0825 {Aluminium oxynitrides} 21/0826 {Silicon aluminium oxynitrides, i.e. sialons} 21/0828 {Carbonitrides or oxycarbonitrides of metals, boron or silicon} 21/083 containing one or more halogen atoms 21/083 Containing one or more halogen with halogens}
21/0494	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one element chosen from metals, silicon or boron are classified in C01B 21/0602 Documents relating to several specific binary compounds are classified in C01B 21/06 only and receive the indexing codes chosen from C01B 21/0602 - C01B 21/076 to identify the specific compounds With two or more other elements chosen from	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 . Hydrazoic acid; Azides; Halogen azides 21/08 . Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 . {Oxynitrides of metals, boron or silicon} 21/0825 {Aluminium oxynitrides} 21/0826 {Silicon aluminium oxynitrides, i.e. sialons} 21/0828 {Carbonitrides or oxycarbonitrides of metals, boron or silicon} 21/083 containing one or more halogen atoms 21/083 {Sinary compounds of nitrogen with halogens} 21/0835 {Nitrogen trifluoride}
21/0494 21/06 21/0602	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one element chosen from metals, silicon or boron are classified in C01B 21/0602 2. Documents relating to several specific binary compounds are classified in C01B 21/06 only and receive the indexing codes chosen from C01B 21/0602 - C01B 21/076 to identify the specific compounds • {with two or more other elements chosen from metals, silicon or boron}	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 . Hydrazoic acid; Azides; Halogen azides 21/082 . Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 . {Oxynitrides of metals, boron or silicon} 21/0825 {Aluminium oxynitrides} 21/0826 {Silicon aluminium oxynitrides, i.e. sialons} 21/0828 {Carbonitrides or oxycarbonitrides of metals, boron or silicon} 21/083 containing one or more halogen atoms 21/083 Containing one or more halogen with halogens}
21/0494 21/06 21/0602 21/0605	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one element chosen from metals, silicon or boron are classified in C01B 21/0602 2. Documents relating to several specific binary compounds are classified in C01B 21/06 only and receive the indexing codes chosen from C01B 21/0602 - C01B 21/076 to identify the specific compounds • {with two or more other elements chosen from metals, silicon or boron} • {Binary compounds of nitrogen with carbon}	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 . Hydrazoic acid; Azides; Halogen azides 21/08 . Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 . {Oxynitrides of metals, boron or silicon} 21/0825 {Aluminium oxynitrides} 21/0826 {Silicon aluminium oxynitrides, i.e. sialons} 21/0828 {Carbonitrides or oxycarbonitrides of metals, boron or silicon} 21/083 containing one or more halogen atoms 21/083 {Sinary compounds of nitrogen with halogens} 21/0835 {Nitrogen trifluoride}
21/0494 21/06 21/0602 21/0605 21/0607	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one element chosen from metals, silicon or boron are classified in C01B 21/0602 2. Documents relating to several specific binary compounds are classified in C01B 21/06 only and receive the indexing codes chosen from C01B 21/0602 - C01B 21/076 to identify the specific compounds • {with two or more other elements chosen from metals, silicon or boron} • {Binary compounds of nitrogen with carbon} • {with alkali metals}	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 . Hydrazoic acid; Azides; Halogen azides 21/082 . Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 . {Oxynitrides of metals, boron or silicon} 21/0825 {Aluminium oxynitrides} 21/0826 {Silicon aluminium oxynitrides, i.e. sialons} 21/0828 {Carbonitrides or oxycarbonitrides of metals, boron or silicon} 21/083 containing one or more halogen atoms 21/0835 {Nitrogen trifluoride} 21/0837 {Purification}
21/0494 21/06 21/0602 21/0605	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one element chosen from metals, silicon or boron are classified in C01B 21/0602 2. Documents relating to several specific binary compounds are classified in C01B 21/06 only and receive the indexing codes chosen from C01B 21/0602 - C01B 21/076 to identify the specific compounds • {with two or more other elements chosen from metals, silicon or boron} • {Binary compounds of nitrogen with carbon}	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 . Hydrazoic acid; Azides; Halogen azides 21/082 . Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 . {Oxynitrides of metals, boron or silicon} 21/0823 {Silicon oxynitrides} 21/0826 {Silicon aluminium oxynitrides, i.e. sialons} 21/0828 {Carbonitrides or oxycarbonitrides of metals, boron or silicon} 21/083 containing one or more halogen atoms 21/083 {Nitrogen trifluoride} 21/0837 {Purification} 21/084 containing also one or more oxygen atoms, e.g.
21/0494 21/06 21/0602 21/0605 21/0607	NOTE In this group, processing steps are indexed with codes chosen from C01B 2210/0001 - C01B 2210/0025 Binary compounds of nitrogen with metals, with silicon, or with boron, {or with carbon, i.e. nitrides; Compounds of nitrogen with more than one metal, silicon or boron}(azides C01B 21/08) NOTES 1. Binary compounds, i.e. compounds of nitrogen with only one other element chosen from metals, silicon, boron or carbon, are classified in groups C01B 21/06 or C01B 21/0605 - C01B 21/076. Compounds of nitrogen with more than one element chosen from metals, silicon or boron are classified in C01B 21/0602 2. Documents relating to several specific binary compounds are classified in C01B 21/06 only and receive the indexing codes chosen from C01B 21/0602 - C01B 21/076 to identify the specific compounds • {with two or more other elements chosen from metals, silicon or boron} • {Binary compounds of nitrogen with carbon} • {with alkali metals}	 21/0726 {Preparation by carboreductive nitridation} 21/0728 {After-treatment, e.g. grinding, purification} 21/076 with titanium or zirconium {or hafnium} 21/0761 {Preparation by direct nitridation of titanium, zirconium or hafnium} 21/0763 {Preparation from titanium, zirconium or hafnium halides} 21/0765 {Preparation by carboreductive nitridation} 21/0766 {Preparation by pyrolysis of nitrogen containing titanium, zirconium or hafnium compounds} 21/0768 {After-treatment, e.g. grinding, purification} 21/08 . Hydrazoic acid; Azides; Halogen azides 21/082 . Compounds containing nitrogen and non-metals {and optionally metals}(C01B 21/06, C01B 21/08 take precedence) 21/0821 . {Oxynitrides of metals, boron or silicon} 21/0825 {Silicon oxynitrides} 21/0826 {Silicon aluminium oxynitrides, i.e. sialons} 21/0828 {Carbonitrides or oxycarbonitrides of metals, boron or silicon} 21/083 containing one or more halogen atoms 21/083 {Binary compounds of nitrogen with halogens} 21/0837 {Purification} 21/084 containing also one or more oxygen atoms, e.g. nitrosyl halides

21/0846	{Nitrosyl parablarate}	21/30	• • • Preparation by oxidation of nitrogen {(C01B 21/26 takes precedence)}
21/0848 21/086	 {Nitrosyl perchlorate} containing one or more sulfur atoms	21/32	Apparatus
21/0865	{Binary compounds of nitrogen with sulfur}	21/32	 Apparatus Nitrogen trioxide (N₂O₃) {(C01B 21/203 takes
21/0803	{Binary compounds of introgen with suntil } containing one or more hydrogen atoms	21/34	precedence)}
	containing one of more hydrogen atoms containing also one or more halogen atoms	21/36	 Nitrogen dioxide (NO₂, N₂O₄) ({C01B 21/203})
21/088 21/09		21,30	C01B 21/26, C01B 21/30 take precedence)
21/09	• • • • {Chloramine, i.e. NH ₂ Cl or dichloramine,	21/38	Nitric acid
21/071	i.e. NHCl ₂ }	21/40	Preparation by absorption of oxides of nitrogen
21/092	• • • containing also one or more metal atoms		$\{(C01B\ 21/26\ takes\ precedence)\}$
21/0923	{Metal imides or amides (silicon imides or	21/42	Preparation from nitrates
21/0/20	amides <u>C01B 21/087</u>)}	21/44	Concentration {(C01B 21/40 takes
21/0926	· · · · { of alkali metals }		precedence)}
21/093	containing also one or more sulfur atoms	21/46	• • • Purification; Separation {; Stabilisation
21/0935	{Imidodisulfonic acid; Nitrilotrisulfonic		$(\underline{\text{C01B 21/40}} \text{ takes precedence})$
	acid; Salts thereof}	21/48	• • Methods for the preparation of nitrates in general
21/094	Nitrosyl containing acids		(particular individual nitrates, <u>see</u> the relevant
21/096	Amidosulfonic acid; Salts thereof		groups of subclasses <u>C01B</u> - <u>C01G</u> , according to
21/097	 containing phosphorus atoms 	21/50	the cation) Nitrous acid; Salts thereof
21/0975	• • {containing also one or more sulfur atoms}	21/30	Nitrous acid, Saits thereof
21/098	Phosphonitrilic dihalides; Polymers thereof	23/00	Noble gases; Compounds thereof (liquefying <u>F25J</u>
21/0983	• • • {Phosphonitrilic difluorides; Polymers		{; noble gases obtained by rectification <u>F25J 3/028</u> })
	thereof}	23/0005	• {Compounds of noble gases}
21/0986	• • • {Phosphonitrilic dichlorides; Polymers	23/001	• {Purification or separation processes of noble
	thereof}		gases}
21/12	Carbamic acid {or thiocarbamic acid}; Salts	23/0015	• • {Chemical processing only}
21/125	thereof	23/0021	• • · {by oxidation}
21/125	{Metal carbamates}	23/0026	• • {by reduction}
21/14	Hydroxylamine; Salts thereof	23/0031	• • · {by complexation}
21/1409	· · · {Preparation}	23/0036	• • {Physical processing only}
21/1418	{by catalytic reduction of nitrogen oxides or	23/0042	• • • {by making use of membranes}
21/1427	nitrates with hydrogen }	23/0047	• • • {characterised by the membrane}
21/1427	• • • • {by reduction of nitrogen oxides or nitrites with bisulfite or sulfur dioxide, e.g. by the	23/0052	• • {by adsorption in solids}
	Raschig process}	23/0057	• • • {characterised by the adsorbent}
21/1436	• • • {by reaction in the gas phase, e.g. of	23/0063	{Carbon based materials}
21/1.50	nitrogen, hydrogen and oxygen}	23/0068	{Zeolites}
21/1445	• • • {of hydoxylamine from its salts}	23/0073	• • • • {Other molecular sieve materials}
21/1454	{of hydroxylamine salts by processes	23/0078	• • • {Temperature swing adsorption}
	not covered by one or more of groups	23/0084	{in getters}
	$\underline{\text{C01B } 21/1418}$ - $\underline{\text{C01B } 21/1445}$, e.g. by	23/0089	 {by absorption in liquids}. {Combined chemical and physical processing}
	conversion of one salt into another}	23/0094	• • {Combined chemical and physical processing}
21/1463	{Concentration}		NOTE
21/1472	{Separation}		In this group, processing steps are
21/1481	{Purification}		indexed with codes chosen from
21/149	{Stabilisation}		<u>C01B 2210/0001</u> - <u>C01B 2210/0025</u>
21/16	Hydrazine; Salts thereof	25/00	Phosphorus; Compounds thereof ({C01B 6/00} ,
21/20	Nitrogen oxides; Oxyacids of nitrogen; Salts thereof	25/00	C01B 21/00, C01B 23/00 take precedence;
21/203	• • {Preparation of nitrogen oxides using a plasma or		perphosphates <u>C01B 15/16</u>)
21/206	an electric discharge \ (No.) (COLD 21/202 + 1	25/003	• {Phosphorus}
21/206	• {Nitric anhydride (N ₂ O ₅) (<u>C01B 21/203</u> takes precedence)}	25/006	• {Stabilisation (<u>C01B 25/04</u> takes precedence)}
21/22	• Nitrous oxide (N ₂ O) {(<u>C01B 21/203</u> takes	25/01	Treating phosphate ores or other raw phosphate
21/22	precedence)}	25/01	materials to obtain phosphorus or phosphorus
21/24	• Nitric oxide (NO) {(C01B 21/203 takes		compounds
21/2-T	precedence)}	25/02	Preparation of phosphorus
21/26	• • Preparation by catalytic {or non-catalytic}	25/023	of red phosphorus
	oxidation of ammonia	25/027	of yellow phosphorus
21/262	• • • {obtaining nitrogen dioxide or tetroxide}	25/04	Purification of phosphorus
21/265	{characterised by the catalyst}	25/043	• of red phosphorus
21/267	{Means for preventing deterioration or loss	25/047	• of yellow phosphorus
	of catalyst or for recovering lost catalyst}	25/06	Hydrogen phosphides
21/28	Apparatus	25/08	• Other phosphides

25/081	• • {of alkali metals, alkaline-earth metals or	25/234 Purification; Stabilisation; Concentration
	magnesium}	(purification concomitant with preparation
25/082	• • {of boron, aluminium, gallium or indium}	C01B 25/22; preparation involving solvent-
25/084	• • • {of boron}	solvent extraction C01B 25/46)
25/085	• • {of aluminium}	25/2343 {Concentration concomitant with
		purification, e.g. elimination of fluorine}
25/087	• • { of gallium or indium }	25/2346 {Concentration}
25/088	• • {containing plural metal}	
25/10	 Halides or oxyhalides of phosphorus 	25/235 Clarification; Stabilisation to prevent post-
25/12	 Oxides of phosphorus 	precipitation of dissolved impurities
25/14	Sulfur, selenium, or tellurium compounds of	25/237 Selective elimination of impurities
	phosphorus	$\{(\underline{\text{C01B } 25/2343} \text{ takes precedence})\}$
25/16		25/2372 {Anionic impurities, e.g. silica or boron
25/16	• Oxyacids of phosphorus; Salts thereof (peroxyacids	compounds}
	or salts thereof <u>C01B 15/00</u>)	25/2375 {Fluoride or fluosilicate anion}
25/161	• • {containing at least one phosphorus atom with an	
	oxidation number less than five, other than those	25/2377 {Sulfate}
	mentioned below; Salts thereof}	25/238 Cationic impurities {, e.g. arsenic
25/163	Phosphorous acid; Salts thereof	compounds}
25/165	Hypophosphorous acid; Salts thereof	25/24 Condensed phosphoric acids
25/168	 Pyrophosphorous acid; Salts thereof 	25/26 . Phosphates (perphosphates C01B 15/16)
		25/265 {General methods for obtaining phosphates}
25/18	Phosphoric acid	
25/185	• • • {Preparation neither from elemental	25/28 Ammonium phosphates
	phosphorus or phosphoric anhydride nor by	25/30 Alkali metal phosphates
	reacting phosphate-containing material with	25/301 {Preparation from liquid orthophosphoric
	an acid, e.g. by reacting phosphate-containing	acid or from an acid solution or suspension
	material with an ion-exchange resin or an acid	of orthophosphates (using ion-exchangers
	salt used alone}	C01B 25/30)}
25/20		25/303 { with elimination of impurities}
25/20	Preparation from elemental phosphorus or	
	phosphoric anhydride	25/305 {Preparation from phosphorus-containing
25/22	• • • Preparation by reacting phosphate-containing	compounds by alkaline treatment}
	material with an acid, e.g. wet process	25/306 {from phosphates}
25/2204	• • • • {Arrangements of vessels used in reacting	25/308 {Methods for converting an alkali
	phosphate-containing material in wet	metal orthophosphate into another one;
	process}	Purification; Decolorasing; Dehydrating;
25/2208	• • • { with an acid or a mixture of acids other than	Drying}
23/2200	sulfuric acid}	25/32 Phosphates of magnesium, calcium, strontium,
25/2212		or barium
25/2212	• • • • • {with hydrochloric acid or hydrogen	
	chloride in aqueous medium}	25/321 {Methods for converting an alkaline earth
25/2216	• • • • { with nitric acid or nitrous vapours in	metal ortho-phosphate into another ortho-
	aqueous medium}	phosphate (by reaction, e.g. of phosphate
25/222	with sulfuric acid, a mixture of acids mainly	rock with phosphoric acid <u>C01B 25/322</u>)}
	consisting of sulfuric acid or a mixture of	25/322 {Preparation by neutralisation of
	compounds forming it in situ, e.g. a mixture	orthophosphoric acid}
	of sulfur dioxide, water and oxygen	25/324 {Preparation from a reaction solution
25/223	only one form of calcium sulfate being	obtained by acidifying with an acid other
23/223		than orthophosphoric acid}
	formed	
25/2235	• • • • • {Anhydrite processes}	25/325 {Preparation by double decomposition}
25/225	Dihydrate process	25/327 { After-treatment (increasing the phosphate
25/226	Hemihydrate process	content of ores $C01B 25/32$)
25/228	one form of calcium sulfate being formed	25/328 {Defluorination during or after the
23/220	and then converted to another form	preparation}
25/2295		25/34 Magnesium phosphates
25/2285	{Dihydrate-anhydrite or hemihydrate-	
	anhydrite process}	25/36 Aluminium phosphates
25/229	• • • • • Hemihydrate-dihydrate process	25/37 Phosphates of heavy metals
25/2295	• • • • • { the conversion being performed	25/372 {of titanium, vanadium, zirconium, niobium,
	in one or more vessels different	hafnium or tantalum}
	from those used for reaction after	25/375 { of iron }
	separation of phosphoric acid}	25/377 { of manganese }
25/231	Dihydrate-hemihydrate process	
		25/38 Condensed phosphates
25/232	Preparation by reacting phosphate	25/385 {of alkaline-earth metals or magnesium}
	containing material with concentrated	25/39 of alkali metals
	sulfuric acid and subsequently lixiviating	25/395 {Preparation and dehydrating}
	the obtained mass, e.g. clinker process	25/40 Polyphosphates
		25/405 {of ammonium}
		22 (or annionan)

25/41	of alkali metals	32/10	• Carbon fluorides, e.g. [CF] _n or [C ₂ F] _n (graphite
25/412	{Preparation from alkali metal	32/10	intercalation thereof C01B 32/22)
23/412	orthophosphates}	32/15	Nano-sized carbon materials
25/414	· · · · · · {Apparatus}	32/152	Fullerenes
25/416	{Pure alkali metal polyphosphates	32/154	Preparation
	from impure starting materials}	32/156	After-treatment
25/418	• • • • • {After-treatment}	32/158	Carbon nanotubes
25/42	Pyrophosphates		NOTE
25/425	• • • • {of alkali metals}		
25/44	Metaphosphates		{In groups C01B 32/158 - C01B 32/18,
25/445	of alkali metals		it is desirable to add indexing codes of C01B 2202/00 - C01B 2202/36 for structural
25/45	containing plural metal, or metal and ammonium		aspects or properties of carbon nanotubes.}
25/451	• • • {containing metal and ammonium}	32/159	single-walled
25/455	• • • containing halogen {(completely halogenated	32/16	Preparation
	alkali metal phosphates CO1D, e.g. lithium	32/162	characterised by catalysts
05/4555	hexafluorophosphate <u>C01D 15/005</u>)}	32/164	involving continuous processes
25/4555	• • • • {Hypochlorite-phosphate double salts, e.g. 4(Na3PO411H2O). NaOCl or so-called	32/166	in liquid phase
	chlorinated trisodium phosphate}	32/168	After-treatment
25/46	Preparation involving solvent-solvent extraction	32/17	Purification
23/40	(solvent extraction in general <u>B01D 11/00</u>)	32/172	Sorting
25/461	• • • { the phosphoric acid present in the medium	32/174	Derivatisation; Solubilisation; Dispersion in
	obtained after reaction being first extracted		solvents
	from the liquid phase formed or separated	32/176	Cutting
	then re-extracted as free acid by using water	32/178	Opening; Filling
	or as a phosphate by using a basic compound	32/18	Nanoonions; Nanoscrolls; Nanohorns;
	(selective extraction of impurities contained in	22/102	Nanocones; Nanowalls
	acid <u>C01B 25/237</u>)}	32/182	Graphene
	<u>NOTES</u>	32/184	Preparation
	1. The extracting agent may be diluted with	32/186	by chemical vapour deposition [CVD]
	a compound or a mixture of compounds	32/188	by epitaxial growth
	which are not solvents for phosphoric acid,	32/19	by exfoliation
	e.g. a hydrocarbon	32/192	starting from graphitic oxides
	2. Documents which belong to more than one	32/194	After-treatment
	subgroup of <u>C01B 25/462</u> - <u>C01B 25/466</u>	32/196	Purification
	are classified by a combination, e.g.	32/198 32/20	Graphene oxide
	<u>C01B 25/462</u> + B4+B8	32/20	. Graphite NOTE
25/462	• • • {the extracting agent being alcohol or a		
	mixture of alcohols}		{In groups <u>C01B 32/20</u> - <u>C01B 32/196</u> ,
25/463	{the extracting agent being a ketone or a		it is desirable to add indexing codes of
05/464	mixture of ketones}		<u>C01B 2204/00</u> - <u>C01B 2204/32</u> for structural aspects or properties of graphene.}
25/464	{the extracting agent being an ether or a mixture of ethers}	22/205	
25/465	{the extracting agent being an ester or a	32/205	. Preparation
	mixture of esters}	32/21	. After-treatment
25/466	• • • { the extracting agent being a nitrogenous	32/215	 Purification; Recovery or purification of graphite formed in iron making, e.g. kish
	solvent or a mixture of nitrogenous solvents		graphite formed in from making, e.g. kish graphite
	such as amines or amides}	32/22	Intercalation
25/467	• • • {the extracting agent being already present	32/225	Expansion; Exfoliation
	during the phosphate-containing material	32/23	Oxidation
25/460	reaction step}	32/25	. Diamond
25/468	• • • {the extraction being performed on the reaction slurry itself, i.e. without separating the acid	32/26	Preparation (by using ultra-high pressure)
	(C01B 25/232 takes precedence)	32/20	<u>B01J 3/06</u> ; by crystal growth <u>C30B 29/04</u>)
		32/28	• After-treatment, e.g. purification, irradiation,
32/00	Carbon; Compounds thereof (C01B 21/00,		separation or recovery
	C01B 23/00 take precedence; percarbonates	32/30	Active carbon
	<u>C01B 15/10</u> ; carbon black <u>C09C 1/48</u>)	32/306	with molecular sieve properties
32/05	Preparation or purification of carbon not covered	32/312	Preparation
	by groups <u>C01B 32/15</u> , <u>C01B 32/20</u> , <u>C01B 32/25</u> ,	32/318	characterised by the starting materials
	<u>C01B 32/30</u>	32/324	from waste materials, e.g. tyres or spent
			sulfite pulp liquor

32/33	from distillation residues of coal or	33/023 by reduction of silica or {free} silica-
32/336	petroleum; from petroleum acid sludge characterised by gaseous activating agents	containing material 33/025 with carbon or a solid carbonaceous material,
32/330	 characterised by gaseous activating agents characterised by non-gaseous activating agents 	i.e. carbo-thermal process
32/348	Metallic compounds	33/027 by decomposition or reduction of gaseous or
32/354	. After-treatment	vaporised silicon compounds other than silica
32/354	Reactivation or regeneration	or silica-containing material
32/366	by physical processes, e.g. by irradiation,	33/029 by decomposition of monosilane
32/300	by using electric current passing through	33/03 by decomposition of silicon halides or
	carbonaceous feedstock or by using	halosilanes or reduction thereof with
	recyclable inert heating bodies	hydrogen as the only reducing agent
32/372	Coating; Grafting; Microencapsulation	33/031 by decomposition of silicon tetraiodide
32/378	Purification	33/033 by reduction of silicon halides or halosilanes
32/382	• • • {Making shaped products, e.g. fibres, spheres,	with a metal or a metallic alloy as the only
	membranes or foam}	reducing agents
32/384	Granulation	33/035 by decomposition or reduction of gaseous or
	NOTE	vaporised silicon compounds in the presence
		of heated filaments of silicon, carbon or a refractory metal, e.g. tantalum or tungsten,
	In this group, the term "granulation" also	or in the presence of heated silicon rods
	<u>covers</u> methods of preparation of active carbon using carbonaceous precursors <u>per se</u>	on which the formed silicon is deposited,
	and binders, e.g. pitch.	a silicon rod being obtained, e.g. Siemens
	and officers, e.g. piten.	process
32/39	 Apparatus for the preparation thereof 	33/037 • Purification (by zone-melting C30B 13/00)
32/40	Carbon monoxide	33/039 by conversion of the silicon into a compound,
32/50	Carbon dioxide	optional purification of the compound, and
32/55	Solidifying	reconversion into silicon
32/60	 Preparation of carbonates or bicarbonates in 	33/04 • Hydrides of silicon
	general (of percarbonates <u>C01B 15/10</u> ; of specific	33/043 • • {Monosilane}
	carbonates or bicarbonates according to the cation	33/046 • • {Purification}
22/50	<u>C01B-C01G</u>)	33/06 • Metal silicides (alloys <u>C22</u>)
32/70	Compounds containing carbon and sulfur, e.g. thiophassess	33/08 • Compounds containing halogen
32/72	thiophosgene . Carbon disulfide	33/10 Compounds containing silicon, fluorine, and other
32/75		elements
32/13	Preparation by reacting sulfur or sulfur compounds with hydrocarbons	33/103 {Fluosilicic acid; Salts thereof}
32/77	Carbon oxysulfide	33/107 • Halogenated silanes
32/80	Phosgene	33/10705 {Tetrafluoride}
32/90	Carbides	33/1071 {Tetrachloride, trichlorosilane or
32/907	Oxycarbides; Sulfocarbides; Mixture of carbides	silicochloroform, dichlorosilane,
32/914	Carbides of single elements	monochlorosilane or mixtures thereof} 33/10715 {prepared by reacting chlorine with silicon or
32/921	Titanium carbide	a silicon-containing material}
32/928	Carbides of actinides	33/10721 {with the preferential formation of
32/935	Carbides of alkali metals, strontium, barium or	tetrachloride}
32,733	magnesium	33/10726 {from silicon}
32/942	Calcium carbide	33/10731 {with the preferential formation of
32/949	Tungsten or molybdenum carbides	trichlorosilane}
32/956	Silicon carbide	33/10736 {from silicon}
32/963	Preparation from compounds containing	33/10742 {prepared by hydrochlorination of silicon or
	silicon	of a silicon-containing material}
32/97	Preparation from SiO or SiO ₂	33/10747 { with the preferential formation of
32/977	Preparation from organic compounds	tetrachloride}
	containing silicon	33/10752 • • • • • {from silicon}
32/984	Preparation from elemental silicon	33/10757 { with the preferential formation of
32/991	Boron carbide	trichlorosilane}
33/00	Silicon; Compounds thereof ({C01B 6/00,}	33/10763 {from silicon}
33/00	C01B 21/00, C01B 23/00 take precedence;	33/10768 {Tetrabromide; Tetraiodide}
	persilicates <u>C01B 15/14</u> ; carbides <u>C01B 32/956</u>)	33/10773 {Halogenated silanes obtained by
33/02	Silicon (forming single crystals or homogeneous)	disproportionation and molecular
23,02	polycrystalline material with defined structure	rearrangement of halogenated silanes}
	C30B)	33/10778 • • • {Purification}
33/021	Preparation (chemical coating from the vapour	33/10784 {by adsorption}
	phase <u>C23C 16/00</u>)	33/10789 {the adsorbing material being formed <u>in</u>
		situ, e.g. by partial hydrolysis}

33/10794	• • • • {by forming addition compounds or	33/163 {by hydrolysis of organosilicon compounds,
	complexes, the reactant being possibly	e.g. ethyl orthosilicate}
	contained in an adsorbent}	33/166 • • • • {by acidification of silicate in the presence of
33/113	• Silicon oxides; Hydrates thereof {(preparing	an inert organic phase}
	monoxide by reduction of siliceous material	33/18 Preparation of finely divided silica neither in
	<u>C01B 33/182</u>)}	sol nor in gel form; After-treatment thereof
33/12	Silica; Hydrates thereof, e.g. lepidoic silicic acid	(preparation of aerogels by dehydrating
33/122	{Lepidoic silicic acid}	gels <u>C01B 33/158</u> ; treatment to enhance the
33/124	• • • {Preparation of adsorbing porous silica not	pigmenting or filling properties <u>C09C</u>)
	in gel form and not finely divided, i.e. silicon	33/181 {by a dry process}
	skeletons, by acidic treatment of siliceous	33/182 {by reduction of a siliceous material,
	materials}	e.g. with a carbonaceous reducing agent
33/126	• • • {Preparation of silica of undetermined type}	and subsequent oxidation of the silicon
33/128	• • • {by acidic treatment of aqueous silicate	monoxide formed}
	solutions}	33/183 {by oxidation or hydrolysis in the vapour
33/14	Colloidal silica, e.g. dispersions, gels, sols	phase of silicon compounds such as
33/141	Preparation of hydrosols or aqueous	halides, trichlorosilane, monosilane}
	dispersions	33/184 {by hydrolysis of tetrafluoride}
33/1412	• • • • {by oxidation of silicon in basic medium}	33/185 {of crystalline silica-polymorphs having
33/1415	• • • • {by suspending finely divided silica in	molecular sieve properties, e.g. silicalites}
	water}	33/186 {from or via fluosilicic acid or salts thereof
33/1417	• • • • {an aqueous dispersion being obtained}	by a wet process}
33/142	• • • • by acidic treatment of silicates	33/187 by acidic treatment of silicates
33/143	of aqueous solutions of silicates	33/193 of aqueous solutions of silicates
33/1435	{using ion exchangers}	33/20 • Silicates (persilicates C01B 15/14 {; containing
33/145	Preparation of hydroorganosols, organosols	aluminium $C01B 33/26$)
33/143	or dispersions in an organic medium	33/22 Magnesium silicates
22/146		33/24 Alkaline-earth metal silicates
33/146	After-treatment of sols ({preparation of hydrosols or aqueous dispersions from	33/26 • Aluminium-containing silicates {, i.e. silico-
	hydroorganosols, organosols or dispersions	aluminates}
	in an organic medium <u>C01B 33/141</u> };	33/28 {Base exchange silicates, e.g. zeolites
	preparation of hydroorganosols, organosols	(regeneration <u>B01J 49/00</u>)}
	or dispersions in an organic medium	33/2807 {Zeolitic silicoaluminates with a
	from hydrosols {or aqueous dispersions}	tridimensional crystalline structure
	C01B 33/145)	possessing molecular sieve properties;
33/1465	• • • • {"Build-up" of particles using only one sol	Isomorphous compounds wherein a part
33/1403	and a "heel" consisting or not of the sol}	of the aluminium ore of the silicon present
33/148	Concentration; Drying; Dehydration;	may be replaced by other elements such
33/110	Stabilisation; Purification {(C01B 33/1465)	as gallium, germanium, phosphorus;
	takes precedence)}	Preparation of zeolitic molecular sieves from
33/1485	• • • • {Stabilisation, e.g. prevention of gelling;	molecular sieves of another type or from
33/1103	Purification}	preformed reacting mixtures}
33/149	· · · · Coating	33/2815 {of type A (UNION CARBIDE trade
33/151	by progressively adding a sol to a different	name; corresponds to GRACE's types Z-12
33/131	sol, i.e. "build-up" of particles using a	or Z-12L)}
	"heel"	33/2823 {from aqueous solutions of an alkali
33/152	Preparation of hydrogels	metal aluminate and an alkali metal
33/1525	{from or via fluosilicic acid or salts	silicate excluding any other source of
33/1323	thereof}	alumina or silica}
33/154	by acidic treatment of aqueous silicate	33/283 {from a reaction mixture containing
33/134	solutions	at least one aluminium silicate or
22/15/12		aluminosilicate of a clay-type, e.g.
33/1543	{using ion exchangers}	kaolin or metakaolin or its exotherm
33/1546	{the first formed hydrosol being	modification or allophane (containing
	converted to a hydrogel by introduction	a single clay substantially chemically
	into an organic medium immiscible or only partly miscible with water}	modified with an acid, i.e. beyond the
22/155		activation state C01B 33/2815)
33/155	Preparation of hydroorganogels or	33/2838 (of faujasite type, or type X or Y (UNION
22/157	organogels	CARBIDE trade names; correspond
33/157	After-treatment of gels	to GRACE's types Z-14 and Z-14HS,
33/158	Purification; Drying; Dehydrating	respectively)}
33/1585	{Dehydration into aerogels}	$33/2846$ {of type X}
33/159	Coating or hydrophobisation	$33/2853$ {of type Y}
33/16	Preparation of silica xerogels	33/2861 (of mordenite type, e.g. ptilolite or
		dachiardite}

33/2869	• • • • {of other types characterised by an X-ray	35/063	• • {Tetrafluoboric acid; Salts thereof}
	spectrum and a definite composition}	35/065	• • • {Tetrafluoboric acid}
33/2876	• • • • {from a reacting mixture containing	35/066	• • • {Alkali metal tetrafluoborates}
	an amine or an organic cation, e.g. a	35/068	• • {Halogenated hydrides}
	quaternary onium cation-ammonium,	35/08	 Compounds containing boron and nitrogen,
22/2004	phosphonium, stibonium}		phosphorus, oxygen, sulfur, selenium or tellurium
33/2884	{ the aluminium or the silicon in the	35/10	Compounds containing boron and oxygen
	network being partly replaced}		(C01B 35/06 takes precedence)
33/2892	{containing an element or a compound	35/1009	• • • {having molecular-sieve properties}
	occluded in the pores of the network, e.g.	35/1018	{Carbonyl compounds derived from boron
	an oxide already present in the starting		hydrides}
22/22	reaction mixture}	35/1027	• • • {Oxides}
33/32	. Alkali metal silicates ({ <u>C01B 33/24</u> },	35/1036	• • • {Boric anhydride}
22/225	C01B 33/26 take precedence)	35/1045	{Oxyacids}
33/325	• • • {After-treatment, e.g. purification or	35/1054	• • • {Orthoboric acid}
	stabilisation of solutions, granulation;	35/1063	{Preparation from boron ores or borates
	Dissolution; Obtaining solid silicate, e.g. from a solution by spray-drying, flashing off water		using acids or salts}
	or adding a coagulant}	35/1072	{by means of ammonia-carbon dioxide}
		35/1081	• • • • {Preparation by working up other natural
	<u>NOTE</u>		sources, e.g. seawater}
	In this group, obtaining solid silicate, e.g.	35/109	• • • • {Purification; Separation; Concentration}
	as a hydrate of a crystalline silicate, from	35/12	• • Borates {(C01B 35/1063 takes precedence)}
	a solution or a hydrate melt by heating or	35/121	{of alkali metal}
	cooling with or without seeding, is not	35/122	{Sodium tetraborates; Hydrates thereof,
	considered as after-treatment, but classified	33/122	e.g. borax}
	in group <u>C01B 33/32</u>	35/123	• • • • • Preparation from boron ores or other
00/04		33/123	borates}
33/36	having base-exchange properties but not having	35/124	• • • • • {Preparation by working up natural
	molecular sieve properties (regeneration thereof	33/124	brines, e.g. seawater}
22/20	<u>B01J 49/00</u>)	35/125	• • • • • {Purification; Concentration;
33/38	Layered base-exchange silicates, e.g. clays,	33/123	Dehydration; Stabilisation; Other after-
	micas or alkali metal silicates of kenyaite		treatment}
	or magadiite type {(activation of naturally occurring clays <u>B01J 20/12</u> ; pillared layered	35/126	• • • { of alkaline-earth metals, beryllium,
	base-exchange silicates B01J 29/049)}	56,120	aluminium or magnesium}
33/40	• • • • Clays	35/127	• • • • {of heavy metals}
33/405	{not containing aluminium}	35/128	{containing plural metal or metal and
33/403		33/120	ammonium}
33/42	Micas {; Interstratified clay-mica products (delaminated mica or vermiculite platelets	35/14	Compounds containing boron and nitrogen,
	obtained by a process involving cation-		phosphorus, sulfur, selenium or tellurium
	exchange C04B 14/208)}	35/143	• • {Phosphates}
33/425	• • • • {not containing aluminium}	35/146	• • • {Compounds containing boron and
33/44	Products obtained from layered base-	20,1.0	nitrogen, e.g. borazoles (ammonium
33/44	exchange silicates by ion-exchange with		tetrafluoborates C01B 35/063; ammonium
	organic compounds such as ammonium,		borates <u>C01B 35/12</u>)}
	phosphonium or sulfonium compounds or		
	by intercalation of organic compounds, e.g.	Compounds	characterised primarily by their physical or
	organoclay material	chemical pro	operties, rather than by their chemical constitution
33/46	Amorphous silicates, e.g. so-called "amorphous	37/00	Compounds having molecular sieve properties but
	zeolites" (crystalline zeolites <u>C01B 39/00</u>)	37700	not having base-exchange properties
25/00	Donor Common de thomas (~ ~ ~ .
35/00	Boron; Compounds thereof (monoborane,		NOTE
	diborane, metal borohydrides or addition complexes thereof C01B 6/00; perborates C01B 15/12; binary		Compounds classified in main group C01B 37/00
	compounds with nitrogen C01B 21/06; {compounds		are also classified in other groups of class C01
	of noble gases C01B 23/0005}; phosphides		according to their composition
	C01B 25/08; carbides C01B 32/991; alloys containing	27/002	(Motellophoophets
	boron C22)	37/002	• {Metallophosphates not containing aluminium, e.g.
35/02	Boron; Borides	27/005	gallophosphates or silicogallophosphates}
35/02	• Boron Bordes • • {Boron}	37/005	• {Silicates, i.e. so-called metallosilicalites or
35/025		37/007	metallozeosilites}
		3 / / (10 /	
33/020	• • {Higher boron hydrides, i.e. containing at least three boron atoms}		. {Borosilicates}
	three boron atoms}	37/02	. Crystalline silica-polymorphs, e.g. silicalites
35/04	three boron atoms} Metal borides	37/02	• Crystalline silica-polymorphs, e.g. silicalites {dealuminated aluminosilicate zeolites}
	three boron atoms}		. Crystalline silica-polymorphs, e.g. silicalites

constitution			
37/06	 Aluminophosphates containing other elements, e.g. metals, boron 	39/16	from aqueous solutions of an alkali metal aluminate and an alkali metal silicate excluding
37/065	• • {the other elements being metals only}		any other source of alumina or silica but seeds
37/08	Silicoaluminophosphates [SAPO compounds] {, e.g. CoSAPO}	39/18	{(C01B 39/145 takes precedence)} from a reaction mixture containing at least one
39/00	Compounds having molecular sieve and base- exchange properties, e.g. crystalline zeolites; Their preparation; After-treatment, e.g. ion-exchange		aluminium silicate or aluminosilicate of a clay type, e.g. kaolin or metakaolin or its exotherm modification or allophane {(C01B 39/145 takes precedence)}
	or dealumination (treatment to modify the sorption	39/20	• • Faujasite type, e.g. type X or Y
	properties, e.g. shaping using a binder, <u>B01J 20/10</u> ; treatment to modify the catalytic properties, e.g. combination of treatments to make the zeolites	39/205	 • Fadjastic type, e.g. type X of T • • {using at least one organic template directing agent; Hexagonal faujasite; Intergrowth products of cubic and hexagonal faujasite}
	appropriate to their use as a catalyst, <u>B01J 29/04</u> ;	39/22	Type X {(C01B 39/205 takes precedence)}
	treatment to improve the ion-exchange properties	39/24	Type Y {(C01B 39/205 takes precedence)}
	<u>B01J 39/14</u>)		
	NOTES	39/26	• Mordenite type {(<u>C01B 39/023</u> , <u>C01B 39/026</u> , <u>C01B 39/06</u> take precedence)}
	 In this group, the following term is used with the meaning indicated: 	39/265	 {using at least one organic template directing agent}
	 "zeolites" means:i. crystalline aluminosilicates with base-	39/28	• • Phillipsite or harmotome type {(C01B 39/023, C01B 39/026, C01B 39/06 take precedence)}
	exchange and molecular sieve properties,	39/30	Erionite or offretite type, e.g. zeolite T
	having three dimensional, microporous	39/305	{using at least one organic template directing
	lattice framework structure of tetrahedral oxide units;		agent}
	ii. compounds isomorphous to those of the	39/32	Type L
	former category, wherein the aluminium or	39/34	Type ZSM-4
	silicon atoms in the framework are partly or	39/36	 Pentasil type, e.g. types ZSM-5, ZSM-8 or ZSM-11
	wholly replaced by atoms of other elements, e.g. by gallium, germanium, phosphorus or	39/365	Type ZSM-8; Type ZSM-11; ZSM 5/11 intermediate}
	boron.	39/38	Type ZSM-5
	2. Compounds classified in main group C01B 39/00 are also classified in other groups of class C01	39/40	• • • using at least one organic template directing agent
	according to their composition	39/42	Type ZSM-12
39/02	 Crystalline aluminosilicate zeolites; Isomorphous compounds thereof; Direct preparation thereof; 	39/44	Ferrierite type, e.g. types ZSM-21, ZSM-35 or ZSM-38
	Preparation thereof starting from a reaction mixture containing a crystalline zeolite of another type, or	39/445	• • { using at least one organic template directing agent}
39/023	from preformed reactants; After-treatment thereof • • {Preparation of physical mixtures or intergrowth}	39/46	Other types characterised by their X-ray diffraction pattern and their defined composition
	products of zeolites chosen from group <u>C01B 39/04</u> or two or more of groups		{(C01B 39/023, C01B 39/026, C01B 39/06 take precedence)}
39/026	C01B 39/14 - C01B 39/48} {After-treatment}	39/48	using at least one organic template directing
39/020		20/50	agent
	agent, e.g. an ionic quaternary ammonium compound or an aminated compound	39/50	 Zeolites wherein inorganic bases or salts occlude channels in the lattice framework, e.g. sodalite, cancrinite, nosean, hauynite {(ultramarine
39/06	. Preparation of isomorphous zeolites characterised		<u>C09C 1/32</u>)}
	by measures to replace the aluminium or silicon	39/52	Sodalites
	atoms in the lattice framework by atoms of other elements {, i.e. by direct or secondary synthesis}	39/54	 Phosphates, e.g. APO or SAPO compounds NOTE
39/065	• • • {Galloaluminosilicates; Group IVB-metalloaluminosilicates; Ferroaluminosilicates}		Phosphates having either a poorly defined or a
39/08	the aluminium atoms being wholly replaced		weak base-exchange capacity such as MAPO's,
39/082	• • • {Gallosilicates}		SAPO's or BAPO's are classified in C01B 37/00
39/085	• • • {Group IVB- metallosilicates}		
39/087	• • • {Ferrosilicates}	-	
39/10	• • • the replacing atoms being {at least} phosphorus atoms	2201/00	Preparation of ozone by electrical discharge
39/12	the replacing atoms being {at least} boron	2201/10	Dischargers used for production of ozone
	atoms	2201/12	. Plate-type dischargers
39/14	Type A	2201/14	Concentric/tubular dischargers
39/145	• • • {using at least one organic template directing	2201/20 2201/22	Electrodes used for obtaining electrical discharge Constructional details of the electrodes
	agent}		

2201/24	Composition of the electrodes	2203/0283	containing a CO-shift step, i.e. a water gas shift
2201/30	Dielectrics used in the electrical dischargers		step
2201/32	Constructional details of the dielectrics	2203/0288	containing two CO-shift steps
2201/34	Composition of the dielectrics	2203/0294	containing three or more CO-shift steps
2201/40	 using several dischargers in series 	2203/04	• containing a purification step for the hydrogen or
2201/50	Part of the product being recycled		the synthesis gas
2201/60	Feed streams for electrical dischargers	2203/0405	Purification by membrane separation
2201/62	Air	2203/041	In-situ membrane purification during hydrogen
2201/64	Oxygen		production
2201/66	Pretreatment of the feed	2203/0415	Purification by absorption in liquids
2201/70	Cooling of the discharger; Means for making	2203/042	Purification by adsorption on solids
	cooling unnecessary	2203/0425	In-situ adsorption process during hydrogen
2201/72	by air		production
2201/74	by liquid	2203/043	Regenerative adsorption process in two or
2201/76	Water		more beds, one for adsorption, the other for
2201/80	Additional processes occurring alongside the		regeneration
	electrical discharges, e.g. catalytic processes	2203/0435	Catalytic purification
2201/82	. Treatment with ultraviolet light	2203/044	Selective oxidation of carbon monoxide
2201/84	. Treatment with magnetic fields	2203/0445	Selective methanation
2201/90	Control of the process	2203/045	Purification by catalytic desulfurisation
2201/90	· Control of the process	2203/0455	Purification by non-catalytic desulfurisation
2202/00	Structure or properties of carbon nanotubes	2203/046	Purification by cryogenic separation
2202/02	Single-walled nanotubes	2203/0465	Composition of the impurity
2202/04	Nanotubes with a specific amount of walls	2203/047	the impurity being carbon monoxide
2202/06	Multi-walled nanotubes	2203/0475	the impurity being carbon dioxide
2202/08	Aligned nanotubes	2203/048	the impurity being an organic compound
2202/10	Filled nanotubes	2203/0485	the impurity being a sulfur compound
2202/20	Nanotubes characterized by their properties	2203/049	the impurity being carbon
2202/22	Electronic properties	2203/0495	the impurity being water
2202/24	Thermal properties	2203/06	Integration with other chemical processes
2202/26	Mechanical properties	2203/061	. Methanol production
2202/28	Solid content in solvents	2203/062	Hydrocarbon production, e.g. Fischer-Tropsch
2202/30	• • Purity	2203/002	process
2202/32	Specific surface area	2203/063	Refinery processes
2202/34	. Length	2203/065	using hydrotreating, e.g. hydrogenation,
2202/36	. Diameter	2203/003	hydrodesulfurisation
		2203/066	with fuel cells
2203/00	Integrated processes for the production of	2203/067	• • • the reforming process taking place in the fuel
	hydrogen or synthesis gas (reactors or details thereof	2202,001	cell
	<u>B01J 2208/00</u> - <u>B01J 2219/00</u>)	2203/068	Ammonia synthesis
2203/02	Processes for making hydrogen or synthesis gas	2203/08	Methods of heating or cooling
2203/0205	containing a reforming step	2203/0805	. Methods of heating the process for making
2203/0211	containing a non-catalytic reforming step	22027 0002	hydrogen or synthesis gas
2203/0216	containing a non-catalytic steam reforming	2203/0811	• • • by combustion of fuel
	step	2203/0816	Heating by flames
2203/0222	containing a non-catalytic carbon dioxide	2203/0822	the fuel containing hydrogen
	reforming step	2203/0827	at least part of the fuel being a recycle stream
	containing a catalytic reforming step	2203/0833	Heating by indirect heat exchange with hot
2203/0233	the reforming step being a steam reforming	2203/0033	fluids, other than combustion gases, product
	step		gases or non-combustive exothermic reaction
2203/0238	• • • • the reforming step being a carbon dioxide		product gases
	c · .		
2202/02/14	reforming step	2203/0838	
2203/0244	the reforming step being an autothermal	2203/0838	by heat exchange with exothermic reactions, other than by combustion of fuel
2203/0244	• • • • the reforming step being an autothermal reforming step, e.g. secondary reforming	2203/0838 2203/0844	by heat exchange with exothermic reactions,
	• • • • the reforming step being an autothermal reforming step, e.g. secondary reforming processes		 by heat exchange with exothermic reactions, other than by combustion of fuel the non-combustive exothermic reaction being another reforming reaction as defined
2203/025	the reforming step being an autothermal reforming step, e.g. secondary reforming processes containing a partial oxidation step		 by heat exchange with exothermic reactions, other than by combustion of fuel the non-combustive exothermic reaction
2203/025 2203/0255	the reforming step being an autothermal reforming step, e.g. secondary reforming processes containing a partial oxidation step containing a non-catalytic partial oxidation step		 by heat exchange with exothermic reactions, other than by combustion of fuel the non-combustive exothermic reaction being another reforming reaction as defined
2203/025 2203/0255	the reforming step being an autothermal reforming step, e.g. secondary reforming processes containing a partial oxidation step containing a non-catalytic partial oxidation step containing a catalytic partial oxidation step	2203/0844 2203/085	by heat exchange with exothermic reactions, other than by combustion of fuel the non-combustive exothermic reaction being another reforming reaction as defined in groups C01B 2203/02 - C01B 2203/0294
2203/025 2203/0255 2203/0261	 the reforming step being an autothermal reforming step, e.g. secondary reforming processes containing a partial oxidation step containing a non-catalytic partial oxidation step containing a catalytic partial oxidation step [CPO] 	2203/0844 2203/085	by heat exchange with exothermic reactions, other than by combustion of fuel the non-combustive exothermic reaction being another reforming reaction as defined in groups C01B 2203/02 - C01B 2203/0294 by electric heating
2203/025 2203/0255 2203/0261 2203/0266	 the reforming step being an autothermal reforming step, e.g. secondary reforming processes containing a partial oxidation step containing a non-catalytic partial oxidation step containing a catalytic partial oxidation step [CPO] . containing a decomposition step 	2203/0844 2203/085 2203/0855	by heat exchange with exothermic reactions, other than by combustion of fuel the non-combustive exothermic reaction being another reforming reaction as defined in groups C01B 2203/02 - C01B 2203/0294 by electric heating by electromagnetic heating
2203/025 2203/0255 2203/0261 2203/0266 2203/0272	 the reforming step being an autothermal reforming step, e.g. secondary reforming processes containing a partial oxidation step containing a non-catalytic partial oxidation step containing a catalytic partial oxidation step [CPO] . containing a decomposition step containing a non-catalytic decomposition step 	2203/0844 2203/085 2203/0855 2203/0861	by heat exchange with exothermic reactions, other than by combustion of fuel the non-combustive exothermic reaction being another reforming reaction as defined in groups C01B 2203/02 - C01B 2203/0294 by electric heating by electromagnetic heating by plasma
2203/025 2203/0255 2203/0261 2203/0266 2203/0272	 the reforming step being an autothermal reforming step, e.g. secondary reforming processes containing a partial oxidation step containing a non-catalytic partial oxidation step containing a catalytic partial oxidation step [CPO] . containing a decomposition step 	2203/0844 2203/085 2203/0855 2203/0861 2203/0866 2203/0872	 by heat exchange with exothermic reactions, other than by combustion of fuel the non-combustive exothermic reaction being another reforming reaction as defined in groups C01B 2203/02 - C01B 2203/0294 by electric heating by plasma by combination of different heating methods
2203/025 2203/0255 2203/0261 2203/0266 2203/0272	 the reforming step being an autothermal reforming step, e.g. secondary reforming processes containing a partial oxidation step containing a non-catalytic partial oxidation step containing a catalytic partial oxidation step [CPO] . containing a decomposition step containing a non-catalytic decomposition step 	2203/0844 2203/085 2203/0855 2203/0861 2203/0866 2203/0872	by heat exchange with exothermic reactions, other than by combustion of fuel the non-combustive exothermic reaction being another reforming reaction as defined in groups C01B 2203/02 - C01B 2203/0294 by electric heating by electromagnetic heating by plasma by combination of different heating methods . Methods of cooling

2203/0888 by evaporation of a fluid		Adjusting the temperature
2203/0894 Generation of steam		Controlling the pressure
2203/10 • Catalysts for performing the hydrogen forming	2203/1633	Measuring the pressure
reactions	2203/1638	Adjusting the pressure
2203/1005 . Arrangement or shape of catalyst	2203/1642	Controlling the product
2203/1011 • • • Packed bed of catalytic structures, e.g.	2203/1647	Controlling the amount of the product
particles, packing elements	2203/1652	Measuring the amount of product
2203/1017 characterised by the form of the structure	2203/1657	the product being hydrogen
2203/1023 Catalysts in the form of a monolith or	2203/1661	the product being carbon monoxide
honeycomb		the product being carbon dioxide
2203/1029 Catalysts in the form of a foam		Controlling the composition of the product
2203/1035 Catalyst coated on equipment surfaces, e.g.		Measuring the composition of the product
reactor walls		
2203/1041 Composition of the catalyst	2203/168	Adjusting the composition of the product
2203/1047 Group VIII metal catalysts		Control based on demand of downstream process
2203/1052 Nickel or cobalt catalysts	2203/169	Controlling the feed
2203/1058 Nickel catalysts	2203/1695	Adjusting the feed of the combustion
2203/1064 Platinum group metal catalysts	2203/80	Aspect of integrated processes for the production
		of hydrogen or synthesis gas not covered by groups
2203/107 Platinum catalysts		<u>C01B 2203/02</u> - <u>C01B 2203/1695</u>
2203/1076 Copper or zinc-based catalysts	2203/82	Several process steps of
2203/1082 Composition of support materials		<u>C01B 2203/02</u> - <u>C01B 2203/08</u> integrated into a
2203/1088 Non-supported catalysts		single apparatus
2203/1094 • • • Promotors or activators	2203/84	Energy production
• Feeding the process for making hydrogen or	2203/86	Carbon dioxide sequestration
synthesis gas	2204/00	Standard or proportion of groupous
2203/1205 Composition of the feed		Structure or properties of graphene
2203/1211 Organic compounds or organic mixtures used	2204/02	. Single layer graphene
in the process for making hydrogen or synthesis	2204/04	Specific amount of layers or specific thickness
gas	2204/06	Graphene nanoribbons
2203/1217 Alcohols	2204/065	characterized by their width or by their aspect
2203/1223 Methanol		ratio
2203/1229 Ethanol	2204/20	Graphene characterized by its properties
2203/1235 Hydrocarbons	2204/22	Electronic properties
2203/1241 Natural gas or methane	2204/24	Thermal properties
2203/1247 Higher hydrocarbons	2204/26	Mechanical properties
		Calida and and in addressed
	2204/28	Solid content in solvents
2203/1252 Cyclic or aromatic hydrocarbons	2204/28 2204/30	
2203/1252 Cyclic or aromatic hydrocarbons2203/1258 Pre-treatment of the feed		Solid content in solvents Purity Size or surface area
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 	2204/30 2204/32	. Purity. Size or surface area
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 	2204/30 2204/32 2210/00	. Purity. Size or surface areaPurification or separation of specific gases
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 Mixing of different feed components 	2204/30 2204/32	. Purity. Size or surface area
2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 Mixing of different feed components 2203/1282 using static mixers	2204/30 2204/32 2210/00	. Purity. Size or surface areaPurification or separation of specific gases
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 Mixing of different feed components 2203/1282 using static mixers 2203/1288 Evaporation of one or more of the different feed 	2204/30 2204/32 2210/00 2210/0001	 Purity Size or surface area Purification or separation of specific gases Separation or purification processing
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 	2204/30 2204/32 2210/00 2210/0001 2210/0003 2210/0004	 Purity Size or surface area Purification or separation of specific gases Separation or purification processing Chemical processing by oxidation
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 Mixing of different feed components 2203/1282 using static mixers 2203/1288 Evaporation of one or more of the different feed 	2204/30 2204/32 2210/00 2210/0001 2210/0003 2210/0004 2210/0006	 Purity Size or surface area Purification or separation of specific gases Separation or purification processing Chemical processing by oxidation by reduction
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 	2204/30 2204/32 2210/00 2210/0001 2210/0003 2210/0004 2210/0006 2210/0007	 . Purity . Size or surface area Purification or separation of specific gases . Separation or purification processing . Chemical processing . by oxidation . by reduction . by complexation
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process 	2204/30 2204/32 2210/00 2210/0001 2210/0003 2210/0004 2210/0006 2210/0007 2210/0009	 . Purity . Size or surface area Purification or separation of specific gases . Separation or purification processing . Chemical processing . by oxidation . by reduction . by complexation . Physical processing
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 . At least two reforming, decomposition or partial 	2204/30 2204/32 2210/00 2210/0001 2210/0003 2210/0004 2210/0006 2210/0007 2210/0009 2210/001	 Purity Size or surface area Purification or separation of specific gases Separation or purification processing Chemical processing by oxidation by reduction by complexation Physical processing by making use of membranes
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 	2204/30 2204/32 2210/00 2210/0001 2210/0003 2210/0004 2210/0007 2210/0009 2210/001 2210/0012	 Purity Size or surface area Purification or separation of specific gases Separation or purification processing Chemical processing by oxidation by reduction by complexation Physical processing by making use of membranes characterised by the membrane
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 . At least two reforming, decomposition or partial 	2204/30 2204/32 2210/00 2210/0001 2210/0004 2210/0006 2210/0007 2210/0009 2210/001 2210/0012 2210/0014	 Purity Size or surface area Purification or separation of specific gases Separation or purification processing Chemical processing by oxidation by reduction by complexation Physical processing by making use of membranes characterised by the membrane by adsorption in solids
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 . At least two reforming, decomposition or partial oxidation steps in parallel 	2204/30 2204/32 2210/00 2210/0001 2210/0004 2210/0006 2210/0007 2210/0009 2210/001 2210/0014 2210/0015	 Purity Size or surface area Purification or separation of specific gases Separation or purification processing Chemical processing by oxidation by reduction by complexation Physical processing by making use of membranes characterised by the membrane by adsorption in solids characterised by the adsorbent
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 . At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 . At least two reforming, decomposition or partial 	2204/30 2204/32 2210/00 2210/0001 2210/0004 2210/0006 2210/0007 2210/0009 2210/001 2210/0012 2210/0015 2210/0017	 Purity Size or surface area Purification or separation of specific gases Separation or purification processing Chemical processing by oxidation by reduction by complexation Physical processing by making use of membranes characterised by the membrane by adsorption in solids characterised by the adsorbent Carbon-based materials
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 . At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 . At least two reforming, decomposition or partial oxidation steps in series 2203/143 Three or more reforming, decomposition or 	2204/30 2204/32 2210/00 2210/0001 2210/0004 2210/0006 2210/0007 2210/0009 2210/001 2210/0012 2210/0015 2210/0017 2210/0018	 Purity Size or surface area Purification or separation of specific gases Separation or purification processing Chemical processing by oxidation by reduction by complexation Physical processing by making use of membranes characterised by the membrane by adsorption in solids characterised by the adsorbent Carbon-based materials Zeolites
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 . At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 . At least two reforming, decomposition or partial oxidation steps in series 2203/143 Three or more reforming, decomposition or partial oxidation steps in series 	2204/30 2204/32 2210/00 2210/0001 2210/0003 2210/0006 2210/0007 2210/0009 2210/001 2210/0012 2210/0014 2210/0015 2210/0018 2210/002	 Purity Size or surface area Purification or separation of specific gases Separation or purification processing Chemical processing by oxidation by reduction by complexation Physical processing by making use of membranes characterised by the membrane by adsorption in solids characterised by the adsorbent Carbon-based materials Zeolites Other molecular sieve materials
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 . At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 . At least two reforming, decomposition or partial oxidation steps in series 2203/143 Three or more reforming, decomposition or partial oxidation steps in series 2203/145 . At least two purification steps in parallel 	2204/30 2204/32 2210/00 2210/0001 2210/0003 2210/0004 2210/0007 2210/0009 2210/001 2210/0012 2210/0014 2210/0015 2210/0017 2210/0018 2210/002 2210/002	. Purity . Size or surface area Purification or separation of specific gases . Separation or purification processing . Chemical processing . by oxidation . by reduction . by complexation . Physical processing . by making use of membranes . characterised by the membrane . by adsorption in solids . characterised by the adsorbent . Carbon-based materials Zeolites . Other molecular sieve materials . Temperature swing adsorption
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 . At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 . At least two reforming, decomposition or partial oxidation steps in series 2203/143 Three or more reforming, decomposition or partial oxidation steps in series 2203/145 . At least two purification steps in parallel 2203/146 . At least two purification steps in series 	2204/30 2204/32 2210/00 2210/0001 2210/0004 2210/0006 2210/0007 2210/0009 2210/001 2210/0014 2210/0015 2210/0017 2210/0018 2210/002 2210/0021 2210/0021	. Purity . Size or surface area Purification or separation of specific gases . Separation or purification processing . Chemical processing . by oxidation . by reduction . by complexation . Physical processing . by making use of membranes . characterised by the membrane . by adsorption in solids . characterised by the adsorbent . Carbon-based materials . Zeolites . Other molecular sieve materials . Temperature swing adsorption . in getters
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 . At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 . At least two reforming, decomposition or partial oxidation steps in series 2203/143 Three or more reforming, decomposition or partial oxidation steps in series 2203/146 . At least two purification steps in series 2203/147 Three or more purification steps in series 	2204/30 2204/32 2210/00 2210/0001 2210/0003 2210/0004 2210/0007 2210/0009 2210/001 2210/0012 2210/0014 2210/0015 2210/0017 2210/0018 2210/002 2210/002	Purity Size or surface area Purification or separation of specific gases . Separation or purification processing Chemical processing by oxidation by reduction by complexation . Physical processing by making use of membranes characterised by the membrane by adsorption in solids characterised by the adsorbent Carbon-based materials Zeolites Other molecular sieve materials Temperature swing adsorption in getters by absorption in liquids
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/14 . At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 . At least two reforming, decomposition or partial oxidation steps in series 2203/143 Three or more reforming, decomposition or partial oxidation steps in series 2203/146 . At least two purification steps in series 2203/147 Three or more purification steps in series 2203/148 . involving a recycle stream to the feed of the 	2204/30 2204/32 2210/00 2210/0001 2210/0004 2210/0006 2210/0007 2210/0009 2210/001 2210/0014 2210/0015 2210/0017 2210/0018 2210/002 2210/0021 2210/0021	. Purity . Size or surface area Purification or separation of specific gases . Separation or purification processing . Chemical processing . by oxidation . by reduction . by complexation . Physical processing . by making use of membranes . characterised by the membrane . by adsorption in solids . characterised by the adsorbent . Carbon-based materials . Zeolites . Other molecular sieve materials . Temperature swing adsorption . in getters
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 . At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 . At least two reforming, decomposition or partial oxidation steps in series 2203/143 Three or more reforming, decomposition or partial oxidation steps in series 2203/145 . At least two purification steps in parallel 2203/147 Three or more purification steps in series 2203/148 . involving a recycle stream to the feed of the process for making hydrogen or synthesis gas 	2204/30 2204/32 2210/00 2210/0001 2210/0004 2210/0006 2210/0007 2210/0009 2210/0012 2210/0014 2210/0015 2210/0017 2210/0018 2210/002 2210/0023 2210/0025 2210/0026	Purity Size or surface area Purification or separation of specific gases . Separation or purification processing Chemical processing by oxidation by reduction by complexation . Physical processing by making use of membranes characterised by the membrane by adsorption in solids characterised by the adsorbent Carbon-based materials Zeolites Other molecular sieve materials Temperature swing adsorption in getters by absorption in liquids
 2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 . At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 . At least two reforming, decomposition or partial oxidation steps in series 2203/143 Three or more reforming, decomposition or partial oxidation steps in series 2203/145 . At least two purification steps in parallel 2203/147 Three or more purification steps in series 2203/148 . involving a recycle stream to the feed of the process for making hydrogen or synthesis gas 2203/16 . Controlling the process 	2204/30 2204/32 2210/00 2210/0001 2210/0004 2210/0006 2210/0007 2210/0009 2210/0012 2210/0014 2210/0015 2210/0017 2210/0018 2210/002 2210/0023 2210/0025 2210/0026	Purity Size or surface area Purification or separation of specific gases . Separation or purification processing Chemical processing by oxidation by reduction by complexation . Physical processing by making use of membranes characterised by the membrane by adsorption in solids characterised by the adsorbent Carbon-based materials Zeolites Other molecular sieve materials Temperature swing adsorption in getters by absorption in liquids . Isotopes of the specific gas
2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 . Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 . Mixing of different feed components 2203/1282 using static mixers 2203/1288 . Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/14 At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 At least two reforming, decomposition or partial oxidation steps in series 2203/143 Three or more reforming, decomposition or partial oxidation steps in series 2203/145 At least two purification steps in parallel 2203/146 At least two purification steps in series 2203/147 Three or more purification steps in series 2203/148 Three or more purification steps in series 2203/149 Three or more purification steps in series 2203/140 Three or more purification steps in series 2203/141 Three or more purification steps in series 2203/142 Three or more purification steps in series 2203/143 Three or more purification steps in series 2203/146 Starting up the process 2203/160 Starting up the process	2204/30 2204/32 2210/00 2210/0001 2210/0004 2210/0006 2210/0007 2210/0009 2210/0012 2210/0014 2210/0015 2210/0017 2210/0018 2210/002 2210/0023 2210/0025 2210/0026	Purity Size or surface area Purification or separation of specific gases . Separation or purification processing Chemical processing by oxidation by reduction by complexation . Physical processing by making use of membranes characterised by the membrane by adsorption in solids characterised by the adsorbent Carbon-based materials Zeolites Other molecular sieve materials Temperature swing adsorption in getters by absorption in liquids . Isotopes of the specific gas . Separation of the specific gas from gas mixtures
2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 Mixing of different feed components 2203/1282 using static mixers 2203/1288 Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 At least two reforming, decomposition or partial oxidation steps in series 2203/143 Three or more reforming, decomposition or partial oxidation steps in series 2203/145 At least two purification steps in parallel 2203/146 At least two purification steps in series 2203/147 Three or more purification steps in series 2203/148 involving a recycle stream to the feed of the process for making hydrogen or synthesis gas 2203/160 Starting up the process 2203/1609 Shutting down the process	2204/30 2204/32 2210/00 2210/0001 2210/0004 2210/0006 2210/0007 2210/0009 2210/0012 2210/0015 2210/0017 2210/0018 2210/002 2210/0021 2210/0023 2210/0026 2210/0028	 Purity Size or surface area Purification or separation of specific gases Separation or purification processing Chemical processing by oxidation by reduction by complexation Physical processing by making use of membranes characterised by the membrane by adsorption in solids characterised by the adsorbent Carbon-based materials Zeolites Other molecular sieve materials Temperature swing adsorption in getters by absorption in liquids Isotopes of the specific gas from gas mixtures containing a minor amount of this specific gas
2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 Mixing of different feed components 2203/1282 using static mixers 2203/1288 Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 At least two reforming, decomposition or partial oxidation steps in series 2203/143 Three or more reforming, decomposition or partial oxidation steps in series 2203/145 At least two purification steps in parallel 2203/146 At least two purification steps in series 2203/147 Three or more purification steps in series 2203/148 involving a recycle stream to the feed of the process for making hydrogen or synthesis gas 2203/1604 Starting up the process 2203/1609 Shutting down the process 2203/1614 Controlling the temperature	2204/30 2204/32 2210/00 2210/0001 2210/0004 2210/0006 2210/0007 2210/0009 2210/0012 2210/0012 2210/0015 2210/0017 2210/0018 2210/002 2210/0021 2210/0023 2210/0028 2210/0029 2210/0029 2210/0031	Purity Size or surface area Purification or separation of specific gases . Separation or purification processing Chemical processing by oxidation by reduction by complexation . Physical processing by making use of membranes characterised by the membrane by adsorption in solids characterised by the adsorbent Carbon-based materials Zeolites Other molecular sieve materials Temperature swing adsorption in getters by absorption in liquids . Isotopes of the specific gas . Separation of the specific gas from gas mixtures containing a minor amount of this specific gas . Obtaining noble gases . Helium
2203/1252 Cyclic or aromatic hydrocarbons 2203/1258 Pre-treatment of the feed 2203/1264 Catalytic pre-treatment of the feed 2203/127 Catalytic desulfurisation 2203/1276 Mixing of different feed components 2203/1282 using static mixers 2203/1288 Evaporation of one or more of the different feed components 2203/1294 Evaporation by heat exchange with hot process stream 2203/14 . Details of the flowsheet 2203/141 At least two reforming, decomposition or partial oxidation steps in parallel 2203/142 At least two reforming, decomposition or partial oxidation steps in series 2203/143 Three or more reforming, decomposition or partial oxidation steps in series 2203/145 At least two purification steps in parallel 2203/146 At least two purification steps in series 2203/147 Three or more purification steps in series 2203/148 involving a recycle stream to the feed of the process for making hydrogen or synthesis gas 2203/160 Starting up the process 2203/1609 Shutting down the process	2204/30 2204/32 2210/00 2210/0001 2210/0004 2210/0006 2210/0007 2210/0009 2210/0012 2210/0012 2210/0015 2210/0017 2210/0018 2210/002 2210/0021 2210/0023 2210/0028 2210/0028	Purity Size or surface area Purification or separation of specific gases . Separation or purification processing Chemical processing by oxidation by reduction by complexation . Physical processing by making use of membranes characterised by the membrane by adsorption in solids characterised by the adsorbent Carbon-based materials Zeolites Other molecular sieve materials Temperature swing adsorption in getters by absorption in liquids . Isotopes of the specific gas . Separation of the specific gas from gas mixtures containing a minor amount of this specific gas . Obtaining noble gases

C01B

2210/0035		Krypton
2210/0037		Xenon
2210/0039		Radon
2210/004		Separation of a mixture of noble gases
2210/0042	N	Taking ultrapure specific gas
2210/0043	Ir	npurity removed
2210/0045		Oxygen
2210/0046		Nitrogen
2210/0048		Air
2210/005		Carbon monoxide
2210/0051		Carbon dioxide
2210/0053		Hydrogen
2210/0054		Hydrogen halides
2210/0056		. Hydrogen fluoride
2210/0057		Hydrogen chloride
2210/0059		. Hydrogen bromide
2210/006		. Hydrogen iodide
2210/0062		Water
2210/0064		Hydrogen sulfide
2210/0065		Ammonia
2210/0067		Hydrogen cyanide
2210/0068		Organic compounds
2210/007		Hydrocarbons
2210/0071		Sulfur oxides
2210/0073		Sulfur halides
2210/0075		Nitrogen oxides
2210/0076		Nitrogen halides
2210/0078		Noble gases
2210/0079		. Helium
2210/0081		. Neon
2210/0082		. Argon
2210/0084		. Krypton
2210/0085		. Xenon
2210/0087		. Radon
2210/0089		Peroxides
2210/009		. Hydrogen peroxide
2210/0092		Ozone
2210/0093		Metals or metal compounds
2210/0095		. Metals
2210/0096		. Metal hydrides
2210/0098		Other impurities
		•