# **CPC** COOPERATIVE PATENT CLASSIFICATION

#### C CHEMISTRY; METALLURGY (NOTES omitted)

#### **CHEMISTRY**

C01 INORGANIC CHEMISTRY (NOTES omitted)

## C01D COMPOUNDS OF ALKALI METALS, i.e. LITHIUM, SODIUM, POTASSIUM,

**RUBIDIUM, CAESIUM, OR FRANCIUM** (metal hydrides {monoborane, diborane or addition complexes thereof} <u>C01B 6/00</u>; salts of oxyacids of halogens <u>C01B 11/00</u>; peroxides, salts of peroxyacids <u>C01B 15/00</u>; sulfides <u>C01B 17/22</u>; thiosulfates, dithionites, polythionates <u>C01B 17/64</u>; compounds containing selenium or tellurium <u>C01B 19/00</u>; binary compounds of nitrogen with metals <u>C01B 21/06</u>; azides <u>C01B 21/08</u>; {compounds other than ammonia and cyanogen, containing nitrogen and other non-metals <u>C01B 21/082</u>}; metal amides <u>C01B 21/092</u>; nitrites <u>C01B 21/50</u>; phosphides <u>C01B 21/50</u>; {compounds of noble gases <u>C01B 23/0005</u>}; phosphides <u>C01B 25/08</u>; salts of oxyacids of phosphorus <u>C01B 25/16</u>; carbides <u>C01B 32/90</u>; compounds containing silicon <u>C01B 33/00</u>; compounds containing boron <u>C01B 35/00</u>; cyanides <u>C01C 3/08</u>; salts of cyanic acid <u>C01C 3/14</u>; salts of cyanamide <u>C01C 3/16</u>; thiocyanates <u>C01C 3/20</u>)

1/00	Oxides or hydroxides of sodium, potassium or alkali metals in general	3/16	<ul> <li>by precipitation or adsorption {(<u>C01D 3/145</u> takes precedence)}</li> </ul>
1/02	• Oxides	3/18	• • with selective solvents
1/04	• Hydroxides	3/20	• • by melting
1/20	• Preparation by reacting oxides or hydroxides with alkali metal salts	3/22	• Preparation in the form of granules, pieces, or other shaped products
1/22	with carbonates or bicarbonates	3/24	. Influencing the crystallisation process
1/24	from or via fluorides or silico-fluorides	3/26	• Preventing the absorption of moisture or caking of
1/26	• Preparation from or via cyano compounds, e.g.		the crystals
	cyanides, cyanamides	5/00	
1/28	• • Purification; Separation	5/00	Sulfates or sulfites of sodium, potassium or alkali metals in general {(sulfites in general C01B 17/62)}
1/30	• • • by crystallisation	5/002	
1/32	• • • by absorption or precipitation	5/002	• {Preventing the absorption of moisture or caking of the crystals by additives}
1/34	with selective solvents	5/004	• {Preparation in the form of granules, pieces or other
1/36	by oxidation	5/004	shaped products}
1/38	• • • by dialysis	5/006	• {Recovery of sodium sulfate from coagulation baths
1/40	by electrolysis	5/000	for the spinning of viscose}
1/42	Concentration; Dehydration	5/008	• {Preparation of potassium sulfate from alunite}
1/44	• Preparation in the form of granules, pieces, or	5/02	<ul> <li>Preparation of sulfates from alkali metal salts and</li> </ul>
	other shaped products	0,02	sulfuric acid or bisulfates; Preparation of bisulfates
3/00	Halides of sodium, potassium or alkali metals in general {(halides in general <u>C01B 9/00</u> )}	5/04	<ul> <li>Preparation of sulfates with the aid of sulfurous acid or sulfites, e.g. Hargreaves process {(pyrosulfites or metabisulfites C01D 5/145)}</li> </ul>
3/02	• Fluorides	5/06	• Preparation of sulfates by double decomposition
3/04	• Chlorides	5/08	• with each other or with ammonium sulfate
3/06	• Preparation by working up brines; seawater or spent lyes	5/10	• with sulfates of magnesium, calcium, strontium, or barium
3/08	• Preparation by working up natural or industrial salt mixtures or siliceous minerals	5/12	• Preparation of double sulfates of magnesium with
3/10	• Bromides	5/14	sodium or potassium
3/12	• Iodides	5/14	<ul> <li>Preparation of sulfites (<u>C01D 5/04</u> takes precedence)</li> </ul>
3/14	• Purification	5/145	• {Pyrosulfites or metabisulfites}
3/145	• • {by solid ion-exchangers or solid chelating	5/145 5/16	
	agents }	5/10	• Purification {(C01D 5/145 takes precedence)}

### C01D

5/18	• Dehydration $\{(\underline{\text{C01D 5/145}} \text{ takes precedence})\}$
7/00	Carbonates of sodium, potassium or alkali metals in general
7/02	• Preparation by double decomposition
7/04	<ul> <li>with a fluoride or silico-fluoride (C01D 1/24)</li> </ul>
	takes precedence)
7/06	• Preparation via sodium or potassium magnesium carbonate
7/07	Preparation from the hydroxides
7/08	• Preparation from or via cyano compounds of sodium or potassium ( <u>C01D 1/26</u> takes precedence)
7/10	• Preparation of bicarbonates from carbonates (ammonia soda process <u>C01D 7/18</u> )
7/12	<ul> <li>Preparation of carbonates from bicarbonates {or bicarbonate-containing product}</li> </ul>
7/123	• • {by thermal decomposition of solids in the absence of a liquid medium}
7/126	• • {Multi-step processes, e.g. from trona to soda ash}
7/14	Preparation of sesquicarbonates
7/16	• Preparation from compounds of sodium or potassium with amines and carbon dioxide
7/18	• Preparation by the ammonia-soda process
,,10	$\{(\underline{\text{C01D } 7/12} \text{ takes precedence})\}$
7/22	• Purification
7/24	Crystallisation
7/26	• • by precipitation or adsorption
7/28	• • with selective solvents
7/30	by oxidation
7/32	by dialysis
7/34	
7/35	<ul><li>by electrolysis</li><li>Varying the content of water of crystallisation or the</li></ul>
1/33	specific gravity {(calcination <u>B01J 6/00, F27B</u> )}
7/27	
7/37	• Densifying sodium carbonate
7/38	• Preparation in the form of granules, pieces or other shaped products
7/40	• Influencing the crystallisation process
7/42 <b>9/00</b>	<ul> <li>Preventing the absorption of moisture or caking</li> <li>Nitrates of sodium, potassium or alkali metals in</li> </ul>
9/00	<b>general</b> {(preparation as fertilizers or of fertilizers containing them C05D 5/00)}
9/02	<ul> <li>Preparation by working-up natural salt mixtures</li> </ul>
9/04	Preparation with liquid nitric acid
9/06	<ul> <li>Preparation with gaseous nitric acid or nitrogen oxides</li> </ul>
9/08	Preparation by double decomposition
9/10	• • with ammonium nitrate
9/12	• • with nitrates or magnesium, calcium, strontium, or barium
9/14	• • of salts of potassium with sodium nitrate
9/16	• Purification
9/18	• Preparation in the form of shaped products, e.g. granules
9/20	• Preventing the absorption of moisture or caking
13/00	Compounds of sodium or potassium not provided for elsewhere
15/00	Lithium compounds
15/00	• {Lithium hexafluorophosphate}
15/005	• {Linnum nexanuorophosphate} • Oxides; Hydroxides
15/02 15/04	. Uxides; Hydroxides . Halides
15/06	• Sulfates; Sulfites

15/08	• Carbonates; Bicarbonates
15/10	• Nitrates
17/00	Rubidium, caesium or francium compounds
17/003	• {Compounds of alkali metals}
17/006	• • {Preparation of potassium compounds comprising
	precipitating potassium ions by an organic
	reagent or extracting them by a liquid organic
	phase }