### CPC COOPERATIVE PATENT CLASSIFICATION

### C CHEMISTRY; METALLURGY

(NOTES omitted)

#### **CHEMISTRY**

# C10 PETROLEUM, GAS OR COKE INDUSTRIES; TECHNICAL GASES CONTAINING CARBON MONOXIDE; FUELS; LUBRICANTS; PEAT

## C10K PURIFYING OR MODIFYING THE CHEMICAL COMPOSITION OF COMBUSTIBLE GASES CONTAINING CARBON MONOXIDE

#### WARNING

In this subclass non-limiting references (in the sense of paragraph 39 of the Guide to the IPC) may still be displayed in the scheme.

1/00	Purifying combustible gases containing carbon	1/106	• • • {containing Fe compounds}
1,00	monoxide (isolation of hydrogen from mixtures	1/107	• • • {containing As-, Sb-, Sn compounds}
	containing hydrogen and carbon monoxide	1/108	{containing Cu compounds}
	C01B 3/50)	1/100	alkaline-reacting {including the revival of the
1/001	• {working-up the condensates (recovering of NH <sub>3</sub>	1/12	used wash liquors }
	and NH <sub>4</sub> salts <u>C01C 1/00</u> ; working-up or purifying	1/121	• • • • {containing NH <sub>3</sub> only (possibly in
	tars and tar-oils $C10C 1/00$ )	1, 121	combination with NH <sub>4</sub> salts)}
1/002	• {Removal of contaminants}	1/122	• • • {containing only carbonates, bicarbonates,
1/003	• • {of acid contaminants, e.g. acid gas removal}		hydroxides or oxides of alkali-metals
1/004	{Sulfur containing contaminants, e.g. hydrogen		(including Mg)}
	sulfide}	1/123	• • • {containing alkali-, earth-alkali- or NH <sub>4</sub> salts
1/005	• • • {Carbon dioxide}		of inorganic acids derived from sulfur}
1/006	{Hydrogen cyanide}	1/124	• • • {containing metal compounds other than
1/007	• • {of metal compounds}		alkali- or earth-alkali carbonates, hydroxides-
1/008	• • • {Alkali metal compounds}		or oxides- or salts of inorganic acids derived
1/02	• Dust removal		from sulfur}
1/022	• • {by baffle plates}	1/125	• • • • {containing Fe compounds}
1/024	• • {by filtration}	1/126	• • • • {containing As-, Sb-, Sn compounds}
1/026	• • {by centrifugal forces (cyclones <u>B04C</u> )}	1/127	• • • • {containing Cu compounds}
1/028	• • {by electrostatic precipitation (separating	1/128	• • • {containing organic oxygen transferring
	dispersed particles from gases or vapour by		compounds, e.g. sulfoxides}
	electrostatic effect in general <u>B03C 3/00</u> )}	1/14	· · · · organic
1/04	<ul> <li>by cooling to condense non-gaseous materials</li> </ul>	1/143	• • • • {containing amino groups}
	$\{(\underline{\text{C10K 1/001}} \text{ takes precedence})\}$	1/146	• • • • {alkali-, earth-alkali- or NH <sub>4</sub> salts}
1/043	• • {adding solvents as vapour to prevent	1/16	• with non-aqueous liquids
1/046	naphthalene- or resin deposits}	1/165	• • • {at temperatures below zero degrees Celsius}
1/046	• • {Reducing the tar content}	1/18	• • hydrocarbon oils {(C10K 1/165 takes
1/06	• combined with spraying with water {(C10K 1/001	1 /20	precedence)}
1 /00	takes precedence)}	1/20	• by treating with solids; Regenerating spent
1/08	<ul> <li>by washing with liquids; Reviving the used wash liquors (gas washers <u>B01D</u>)</li> </ul>		purifying masses {(separation by adsorption B01D 53/02; separation by chemical reaction
1/085	• • {two direct washing treatments, one with an		B01D 53/34; refining of hydrocarbon oils with acids
1/063	aqueous liquid and one with a non-aqueous		C10G 17/02, C10G 27/02, C10G 29/12)}
	liquid}	1/205	• • {Methods and apparatus for treating the purifying
1/10	with aqueous liquids {(alkaline reacting aqueous)	1,200	masses without their regeneration (recovering
1, 10	liquids C10K 1/12)}		of sulfur <u>C01B 17/00</u> ; recovering of cyanide
1/101	• • • {with water only}		compounds <u>C01C 3/00</u> )}
1/102	• • {containing free acid}	1/22	<ul> <li>Apparatus, e.g. dry box purifiers</li> </ul>
1/103	• • • {alkali- or earth-alkali- or NH <sub>4</sub> salts or	1/24	Supporting means for the purifying material
-,	inorganic acids derived from sulfur}	1/26	Regeneration of the purifying material {contains
1/105	• • • {containing metal compounds other than alkali-		also apparatus for the regeneration of the
	or earth-alkali carbonates, -hydroxides, oxides,		purifying material}
	or salts of inorganic acids derived from sulfur}	1/28	Controlling the gas flow through the purifiers

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1/30 1/32 1/34	<ul> <li>with moving purifying masses</li> <li>with selectively adsorptive solids, e.g. active carbon</li> <li>by catalytic conversion of impurities to more readily removable materials</li> </ul>		
3/00	Modifying the chemical composition of combustible gases containing carbon monoxide to produce an improved fuel, e.g. one of different calorific value, which may be free from carbon monoxide		
3/001	• {by thermal treatment}		
3/003	• • {Reducing the tar content}		
3/005	• • • {by partial oxidation}		
3/006	• • {by steam reforming}		
3/008	• • {by cracking}		
3/02	<ul> <li>by catalytic treatment</li> </ul>		
3/023	• • {Reducing the tar content}		
3/026	<ul> <li>{Increasing the carbon monoxide content, e.g. reverse water-gas shift [RWGS]}</li> </ul>		
3/04	<ul><li>reducing the carbon monoxide content {, e.g. water-gas shift [WGS]}</li></ul>		
3/06	• by mixing with gases		

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